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(11)

EP 1 241 191 A1

(12)

EUROPEAN PATENT APPLICATION published in accordance with Art. 158(3) EPC

- (43) Date of publication: 18.09.2002 Bulletin 2002/38
- (21) Application number: 01930155.5
- (22) Date of filing: 15.05.2001

- (51) Int CI.7: **C08F 210/02**, C08L 23/08, C08L 23/10
- (86) International application number: PCT/JP01/04038
- (87) International publication number:

 WO 01/088000*(22.11.2001 Gazette 2001/47)

- (84) Designated Contracting States: **DE FR GB IT NL**
- (30) Priority: 16.05.2000 JP 2000148646 12.07.2000 JP 2000211620 27.07.2000 JP 2000226632
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- (54) ETHYLENE COPOLYMER AND ETHYLENE/\$g(a)-OLEFIN COPOLYMER COMPOSITION AND PROPYLENE POLYMER COMPOSITION BOTH CONTAINING THE ETHYLENE COPOLYMER

(57) The present invention relates to an ethylene-based copolymer useful as a modifier for resins, e.g., polypropylene, and an ethylene/ α -olefin copolymer composition containing the same ethylene-based copolymer.

The ethylene-based copolymer [I] of the present invention is characterized by:

(a-1) having temperature (T_m) at the maximum peak in the endothermic curve, measured by a differential scanning calorimeter (DSC), of 40 to 90° C, (a-2) containing the component soluble in decane at normal temperature at 1 to 70% by weight, and (a-3) containing the component soluble in decane at normal temperature which comprises (i) a recurring unit derived from ethylene and (ii) a recurring unit derived from an α -olefin of 4 or more carbon atoms, and

in which a content of (i) the recurring unit derived from ethylene is 50 to 75% by mole. The ethylene-based co-polymer composition [II] of the present invention is characterized by containing the ethylene-based copolymer

which satisfies the following conditions (b-1) to (b-3):

(b-1) having temperature (T_m) at the maximum peak in the endothermic curve of the component soluble in decane at 64°C, measured by a differential scanning calorimeter (DSC), of 40 to 90°C,

(b-2) containing the component soluble in decane at 64°C which contains the component soluble in decane at normal temperature at 1 to 70% by weight, and

(b-3) containing the component soluble in decane at normal temperature which comprises (i) a recurring unit derived from ethylene and (ii) a recurring unit derived from an α -olefin of 4 or more carbon atoms, and

in which a content of (i) the recurring unit derived from ethylene is 50 to 75% by mole.

Description

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FIELD OF THE INVENTION

[0001] The present invention relates to an ethylene-based copolymer useful as a modifier for resins, e.g., polypropylene, and an ethylene/ α -olefin copolymer composition containing the same ethylene-based copolymer. The present invention also relates to a propylene-based copolymer composition containing the same ethylene/ α -olefin copolymer composition, and its use.

10 BACKGROUND OF THE INVENTION

[0002] It is well known that a polypropylene resin is blended with an elastomer, e.g., ethylene/propylene or ethylene/butene copolymer, to improve its impact resistance.

[0003] However, blend of the elastomer led deterioration in rigidity of the polypropylene resin, therefore there is a limit in the amount of the blended elastomer. A formed resin article, e.g., that of polypropylene resin, may be required to have impact resistance both at normal temperature and low temperature. Impact resistance at low temperature may not always correspond to that at normal temperature. Blend of a soft rubber as the modifier may be one method for improving impact resistance at low temperature. However, blend of the soft rubber with a polypropylene resin leads rigidity of the formed article to deteriorate and causes the same problem as above.

[0004] Therefore, the modifier capable of keeping the resin in which it is blended well balanced with respect to rigidity and impact resistance has been in demand. On the other hand, a formed resin article, e.g., that described above, is also required to be difficult to fracture under service conditions, by which is meant that it is required to have a high tensile breaking point strength and impact strength while keeping a high rigidity. In other words, the resin modifier capable of keeping the resin in which it is blended well balanced with respect to rigidity and tensile breaking elongation has been also in demand.

[0005] Japanese Patent Laid-Open Publication No.192500/1994 discloses a composition having well-balanced rigidity and impact resistance by incorporating a polypropylene polymer with an ethylene/ α -olefin copolymer of specific properties. However, this composition is insufficient in the balances between rigidity and impact strength at low temperature and between rigidity and tensile elongation properties, and is required to be further improved.

[0006] Moreover, when modifying resin, some resin modifiers for the above-described resins cause problems, e.g., blocking during the kneading process at a supply section of the kneader. e.g., hopper, to deteriorate productivity and unstable properties of the modified resin.

[0007] Therefore, development of the resin modifier capable of improving properties of polypropylene or other resins, in particular tensile breaking point elongation and impact resistance at low temperature, while keeping its rigidity, and, at the same time, causing little problems, e.g., blocking, and excellent in productivity and workability has been keenly demanded.

[0008] The inventors of the present invention have already proposed specific compositions of [A] a copolymer of ethylene and α -olefin of 4 to 20 carbon atoms, [B] a copolymer of ethylene and α -olefin of 3 to 20 carbon atoms and [C-1] a high-density polyethylene, as the modifiers which can satisfy the above requirements (Japanese Patent Laid-Open Publication No.273563/1999. However, these compositions cannot always give the modified resin of well-bal-anced impact resistance and bending modulus of elasticity.

DISCLOSURE OF THE INVENTION

45 [0009] The ethylene-based copolymer [I] of the present invention is characterized by:

(a-1) having temperature (T_m) at the maximum peak in the endothermic curve, measured by a differential scanning calorimeter (DSC), of 40 to 90°C,

(a-2) containing the component soluble in decane at normal temperature at 1 to 70% by weight, and

(a-3) containing the component soluble in decane at normal temperature which comprises (i) a recurring unit derived from ethylene and (ii) a recurring unit derived from an α -olefin of 4 or more carbon atoms, and

in which a content of (i) the recurring unit derived from ethylene is 50 to 75 % by mole.

[0010] The ethylene-based copolymer composition [II] of the present invention is characterized by containing the ethylene-based copolymer-which satisfies the following conditions (b-1) to (b-3):

(b-1) having temperature (T_m) at the maximum peak in the endothermic curve of the component soluble in decane at 64°C, measured by a differential scanning calorimeter (DSC), of 40 to 90°C,

(b-2) containing the component soluble in decane at 64°C which contains the component soluble in decane at normal temperature at 1 to 70% by weight, and

(b-3) containing the component soluble in decane at normal temperature which comprises (i) a recurring unit derived from ethylene and (ii) a recurring unit derived from an α -olefin of 4 or more carbon atoms, and

in which a content of (i) the recurring unit derived from ethylene is 50 to 75 % by mole.

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[0011] The propylene-based copolymer composition [III] of the present invention is characterized by containing the propylene-based copolymer and satisfying the following conditions (c-1) to (c-3):

(c-1) containing the component soluble in decane at 64° C at 1 to 99% by weight which has a temperature (T_{m}) at the maximum peak in the endothermic curve, measured by a differential scanning calorimeter (DSC), of 40 to 90°C, (c-2) containing the component soluble in decane at 64°C which contains the component soluble in decane at normal temperature at 1 to 70% by weight, and

(c-3) containing the component soluble in decane at normal temperature which comprises (i) a recurring unit derived from ethylene and (ii) a recurring unit derived from an α -olefin of 4 or more carbon atoms, and

in which a content of (i) the recurring unit derived from ethylene is 50 to 75 %.by mole.

[0012] The ethylene/ α -olefin copolymer composition of the present invention comprises: 1 to 70% by weight of [A] a copolymer of ethylene and α -olefin of 4 to 20 carbon atoms which comprises (A-i) 50 to 70% by mole of (a) a recurring unit derived from ethylene and 30 to 50% by mole of (b) a recurring unit derived from an α -olefin of 4 to 20 carbon atoms, and 30 to 99% by weight of [B] an at least one of ethylene-based copolymer comprising (a) ethylene, (b) at least one monomer selected form α -olefin of 4 to 20 carbon atoms and cyclic olefin-based compound, and having (B-i) a density of 0.870 to 0.895g/cm³.

[0013] It is preferable that the ethylene/ α -olefin copolymer composition have a melt flow rate (MFR at 190°C and a load of 2.16kg) of 0.01 to 50g/10 minutes and that ratio of density (d_b) of the ethylene-based copolymer [B] to density (d_a) of the ethylene/ α -olefin copolymer [A] (d_b/d_a) of 1.05 or less. It is also preferable that the above-described ethylene/ α -olefin copolymer [A] has (A-ii) a glass transition temperature of -60°C or lower, determined by a differential scanning calorimeter (DSC). crystallinity of 1% or less, and (A-iii) B value of 0.9 to 1.5, given by the following general formula (I):

B value =
$$[P_{0E}]/(2[P_E][P_0])$$
 (1)

wherein. $[P_E]$ is molar fraction of the ethylene-derived recurring unit in the copolymer, $[P_0]$ is molar fraction of the α -olefin-derived recurring unit in the copolymer, and $[P_{0E}]$ is a number ratio of the ethylene/ α -olefin copolymer chains to the total dyad chains in the copolymer, determined by the ¹³C-NMR spectroscopy.

[0014] It is also preferable that the above-described ethylene/ α -olefin copolymer [A] have (A-iv) an intrinsic viscosity [η] of 0.1 to 10.0d/g determined in decalin at 135°C.

[0015] It is also preferable that the ethylene-based copolymer [B] for the ethylene/ α -olefin copolymer composition of the present invention have a melt flow rate (at 190°C and a load of 2.16kg) of 0.1 to 50g/10 minutes and (B-iii) temperature (T_m) at the maximum peak in the endothermic curve, measured by a differential scanning calorimeter (DSC), be correlated with density (d) by the following relationship:

Tm < 400 × d-250

[0016] The above ethylene/ α -olefin copolymer composition is suitable for producing a resin modifier.

[0017] The method of the present invention for modifying resin is characterized by blending the pellets of the above ethylene/ α -olefin copolymer composition in the molten resin it tries to modify.

[0018] The propylene-based polymer composition (1) of the present invention is comprised of the propylene-based polymer [C-1], ethylene/ α -olefin copolymer [A] and ethylene-based copolymer [B], and characterized by:

- (i) containing the propylene-based polymer [C-1] at 99 to 1% by weight, and ethylene/ α -olefin copolymer [A] and ethylene-based copolymer [B] at 1 to 99% by weight (total content of [A] and [B]), and
- (ii) having a content ratio of the ethylene/α-olefin copolymer [A] to the ethylene-based copolymer [B], i.e., [A]/[B] content-ratio.-is-1/99-to-70/30.

[0019] It is preferable that the propylene-based polymer [C-1] be contained at 98 to 60% by weight, and the ethylene/

 α -olefin copolymer [A] and ethylene-based copolymer [B] at 2 to 40% by weight (total content of [A] and [B]).

[0020] It is also preferable that the propylene-based polymer [C-1] have a melt flow rate (at 230°C and a load of 2.16kg) of 0.01g/10 minutes or more.

[0021] It is preferable that the ethylene/ α -olefin copolymer of the present invention have a damping factor peak caused from the glass transition temperature of the propylene-based polymer [C-1] and that caused from the glass transition temperature of the ethylene/ α -olefin copolymer composition (composed of the ethylene/ α -olefin copolymer [A] and ethylene-based copolymer [B]), when the temperature-dependence of the modulus of elasticity is measured, and that these peaks are separated from each other.

[0022] The propylene-based copolymer composition (2) of the present invention is comprised of 20 to 93% by weight of [C-2], 6 to 79% by weight of [AB] and 1 to 25% by weight of [D]:

[C-2]: a propylene-based polymer:

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- (1) having a melt flow rate (230°C and a load of 2.16kg) of 0.1 to 400g/10 minutes, determined in accordance with ASTM D-1238,
- (2) containing 0.01 to 30% by weight of the component soluble in decane at normal temperature, and an intrinsic viscosity $[\eta]$ of 0.2 to 10dl/g, determined in decalin at 135°C, and
- (3) having a pentad isotacticity (I_5) of 0.95 or more, determined by the ¹³C-NMR spectroscopy for the component insoluble in decane at normal temperature;

[AB]: an ethylene/α-olefin copolymer composition, composed of:

[A-1]: an ethylene/ α -olefin copolymer of ethylene and α -olefin of 4 to 20 carbon atoms, (i) comprising 50 to 70% by mole of (a) a recurring unit derived from the ethylene and 30 to 50% by mole of (b) a recurring unit derived from the α -olefin of 4 or more carbon atoms, and (ii) having an intrinsic viscosity [η] of 0.1 to 10.0d/g, determined in decalin at 135°C, and

[B-1]: an ethylene-based copolymer of (a) ethylene and (b) at least one compound selected from the group consisting of α -olefins and cyclo-olefins of 3 to 20 carbon atoms, having (i) a density of 0.870 to 0.895g/cm³ and (ii) a melt flow rate (190°C and a load of 2.16kg) of 0.1 to 50g/10 minutes,

wherein the copolymers [A-1] and [B-1] are incorporated at 1 to 50% and 50 to 99% by weight, respectively, based on the whole composition, and

[D]: an inorganic filler.

[0023] It is preferable that ratio of density (d_2) of the ethylene-based copolymer [B-1] to density (d_1) of the ethylene/ α -olefin copolymer [A-1] (d_2/d_1) be 1.05 or less.

[0024] It is preferable that the propylene-based copolymer composition (2) of the present invention have a damping factor peak derived from the glass transition temperature of the propylene-based polymer [C-2] and that derived from the glass transition temperature of the ethylene/ α -olefin composition [AB], when the temperature-dependence of the modulus of elasticity is measured, and that these peaks be separated from each other.

[0025] The propylene-based copolymer composition (2) is used for the formed articles, suitable for sheets and injection-molded articles, e.g., automobile interiors and exteriors, and electric appliance cases.

[0026] The propylene-based copolymer composition (3) of the present invention is composed of 20 to 95% by weight of [C-3] and 5 to 80% by weight of [AB-1]:

Wherein [C-3] is a propylene/α-olefin copolymer, and

- (1) having an intrinsic viscosity [η] of 0.01 to 10dl/g, determined in decalin at 135°C, and
- (2) comprising propylene and 1.5 to 10% by mole of an α -olefin. other than propylene, of 2 to 20 carbon atoms, and
- [AB-1] is an ethylene/ α -olefin copolymer composition, composed of:
 - [A-1]: an ethylene/ α -olefin copolymer of ethylene and α -olefin of 4 to 20 carbon atoms,
 - (i) comprising 50 to 70% by mole of (a) a recurring unit derived from the ethylene and 30 to 50% by mole -of-(b)-a recurring unit derived from the α-olefin of 4 to 20 carbon atoms, and
 - (ii) having an intrinsic viscosity [η] of 0.1 to 10.0dl/g, determined in decalin at 135°C, and
 - [B-1]: an ethylene-based copolymer of (a) ethylene and (b) at least one compound selected from the group

consisting of α -olefins and cyclo-olefins of 3 to 20 carbon atoms, having

- (i) a density of 0.870 to 0.895g/cm³ and
- (ii) a melt flow rate (at 190°C and a load of 2.16kg) of 0.1 to 50g/10 minutes,

wherein the copolymers [A-1] and [B-1] are blended at 1 to 50% and 50 to 99% by weight, respectively, based on the whole composition.

[0027] It is preferable for the propylene/ α -olefin copolymer composition (3) that ratio of density (d₂) of the ethylene-based copolymer [B-1] to density (d₁) of the ethylene/ α -olefin copolymer [A-1] (d₂/d₁) be 1.05 or less.

[0028] It is preferable that the propylene/ α -olefin copolymer composition (3) have a damping factor peak derived from the glass transition temperature of the propylene/ α -olefin [C-3] and that derived from the glass transition temperature of the ethylene/ α -olefin copolymer [AB-1], when the temperature-dependence of the modulus of elasticity is measured, and that these peaks be separated from each other.

[0029] It is preferable that the propylene/ α -olefin copolymer [C-3] has a microisotacticity of 0.8 or more, with respect to the triad chains of propylene.

[0030] It is also preferable that the propylene/ α -olefin copolymer [C-3] has a molecular weight distribution (Mw/Mn) of 6 or less, determined by GPC and glass transition temperature Tg of -10°C or lower.

[0031] The propylene/ α -olefin copolymer composition is used for the formed articles, suitable for sheets, films and, in particular, filaments. It is also suitable for medical containers and tubes, and transfusion containers and tubes.

BEST MODE FOR CARRYING OUT THE INVENTION

[0032] The novel copolymer and copolymer composition of the present invention and use thereof are described more concretely.

[0033] First, the novel ethylene-based copolymer [I] of the present invention is described.

[Ethylene-based copolymer [I]]

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[0034] The ethylene-based copolymer [I] of the present invention comprises ethylene and α -olefin other than ethylene, containing the recurring unit derived from ethylene and that derived from α -olefin other than ethylene. It may contain the recurring unit derived from a cyclo-olefin.

[0035] The concrete examples of the α -olefins, other than ethylene, useful for the present invention include propylene, 1-butene, 1-pentene, 1-hexene, 3-methyl-1-butene, 3-methyl-1-pentene, 3-ethyl-1-pentene, 4-methyl-1-pentene, 4-methyl-1-hexene, 1-octene, 3-ethyl-1-hexene, 1-decene, 1-dodecene, 1-tetradecene, 1-hexadecene, 1-octadecene, and eicosene, of which 1-butene, 1-hexene and 1-octene are more preferable.

[0036] The cyclo-olefins useful for the present invention include those represented by the following general formula (1-1) or (1-2).

[0037] In the above formula, "n" is 0 or 1; "m" is 0 or a positive integer; "k" is 0 or 1, where the ring involving "k" is 6-membered when "k" is 1 and 5-membered when "k" is 0; and R¹ to R¹8 and Rª and R⁰ in the formula (1-1) are each H.,a.halogen_atom.(F,_CI, Br.or I) or hydrocarbon-group. Further, in-the-formula, R¹5-and-R¹6, R¹7 and R¹8, R¹5 and R¹8, or R¹6 and R¹7, may be bound to each other to form, in a concerted manner, a monoor poly-cycle group which may contain a double bond.

[0038] In the above formula (1-2), "p" and "q" are each 0 or a positive integer; "r" and "s" are each 0, 1 or 2; and R²¹ to R³⁹ are each H. a halogen atom, or hydrocarbon or alkoxy group.

[0039] The halogen atom in the formula (1-2) is the same as that in the formula (1-1).

[0040] The examples of the hydrocarbon groups normally include an alkyl group of 1 to 20 carbon atoms, and cycloalkyl group and aromatic hydrocarbon group of 3 to 15 carbon atoms. More concretely, the alkyl groups of 1 to 20 carbon atoms include methyl, ethyl, propyl, isopropyl, amyl, hexyl, octyl, decyl, dodecyl, and octadecyl. These groups may be substituted with a halogen atom.

[0041] The cycloalkyl groups include cyclohexyl; and aromatic hydrocarbon groups include aryl and aralkyl, more concretely phenyl, tolyl, naphthyl, benzyl and phenylethyl.

[0042] The alkoxy groups include methoxy, ethoxy and propoxy, wherein the carbon atom to which R^{29} and R^{30} are bound may be bound to the carbon atom to which R^{33} or R^{31} is bound, either directly or via an alkylene group of 1 to 3 carbon atoms. When these carbon atoms are bound to each other via an alkylene group, R^{29} and R^{30} or R^{30} and R^{31} jointly form an alkylene group in methylene (CH₂), ethylene (CH₂CH₂) or propylene (CH₂CH₂CH₂) group.

[0043] Moreover. R³⁵ and R³² or R³⁵ and R³⁹ may be bound to each other to form a mono- or poly-cyclic aromatic ring, when r=s=0. The concrete examples of the cyclo-olefins represented by the general formula (1-1) or (1-2) include a bicyclo-2-heptane derivative (bicyclohept-2-ene-2 derivative), tricyclo-3-decene derivative, tricyclo-3-undecene derivative, tetracyclo-3-dodecene derivative, pentacyclo-4-pentadecene derivative, pentacyclopentadecadiene derivative, pentacyclo-3-pentadecene derivative, hexacyclo-4-heptadecene derivative, heptacyclo-5-eicosene derivative, heptacyclo-4-eicosene derivative, heptacyclo-5-heneicosene derivative, octacyclo-5-docosene derivative, nonacyclo-5-pentacosene derivative, nonacyclo-6-hexacosene derivative. cyclopentadiene-acenaphthylene adduct, 1,4-metano-1,4,4a,9a-tetrahydrofluorene derivative, and 1,4-metano-1.4.4a.5.10.10a-hexahydroanthracene derivative.

[0044] The ethylene-based copolymer [I] of the present invention preferably has an MFR (190°C and a load of 2.16kg) of 0.01 to 50g/10 minutes, more preferably 0.01 to 30g/10 minutes.

(1) Component soluble in decane at normal temperature

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[0045] The ethylene-based copolymer [I] of the present invention contains the component soluble in decane at normal temperature at 1 to 70% by weight, preferably 1 to 60%, more preferably 1 to 50%, still more preferably 1 to 29%. wherein normal temperature means room temperature, normally 23°C.

[0046] The component soluble in decane at normal temperature is comprising the recurring units derived from (i) ethylene and (ii) an α -olefin of 4 or more carbon atoms.

[0047] The component soluble in decane at normal temperature contains the ethylene-derived recurring unit at 50 to 75% by mole, preferably 55 to 75%, more preferably 60 to 72%, still more preferably 60 to 70%. The balance comprises the unit derived from an α -olefin of 4 or more carbon atoms.

[0048] The component soluble in decane at normal temperature may contain the above-described cyclo-olefin, propylene or other compounds at 3% by mole or less as the total content, in addition to the ethylene-derived recurring unit.

[0049] The decane-soluble component preferably has a glass transition temperature of -60°C or lower, determined

by-a differential scanning calorimeter (DSC), more preferably -65°C or lower, and preferably crystallinity of 1% or less, preferably 0.5% or less

[0050] The decane-soluble component preferably has an intrinsic viscosity $[\eta]$ of 0.1 to 10.0dl/g, determined in decalin at 135°C, more preferably 1 to 8dl/g, still more preferably 2 to 7dl/g.

[0051] The decane-soluble component preferably has a B value of 0.9 to 1.5, more preferably 1.0 to 1.2, wherein the B value is given by the following general formula (1):

B value =
$$[P_{0E}]/(2[P_E][P_0])$$
 (1)

wherein. $[P_E]$ is molar fraction of the ethylene-derived recurring unit in the copolymer, $[P_0]$ is molar fraction of the α -olefin-derived recurring unit in the copolymer, and $[P_{0E}]$ is a number ratio of the ethylene/ α -olefin copolymer chains to the total dyad chains in the copolymer, determined by the ¹³C-NMR spectroscopy.

[0052] The B value is an index representing the distributions of ethylene and an α -olefin of 4 to 20 carbon atoms, and can be determined by the methods proposed by, e.g., J.C. Randall (Macromolecules, 15, 353 (1982)) and J. Ray (Macromolecules, 10, 773 (1977)).

[0053] As the B value increases, the block chains in the ethylene-based copolymer become shorter, making the ethylene and α -olefin distributions more uniform, and copolymer rubber composition distribution narrower. As the B value decreases to below 1.0, the ethylene-based copolymer composition distribution becomes broader, possibly deteriorating its handling characteristics.

[0054] It is preferable that ratio of density (d₃) of the component soluble in decane at normal temperature to density (d_A) of the ethylene-based copolymer per $se(d_B/d_A)$ be 1.01 to 1.05, preferably 1.01 to 1.04.

[0055] The ethylene-based copolymer having the (d_B/d_A) ratio in the above range has excellent transparency and breaking elongation, and exhibits excellent properties as a modifier.

(2) Component insoluble in decane at normal temperature

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[0056] The component insoluble in decane at normal temperature of the ethylene-based copolymer of the present invention contains the recurring unit derived from ethylene and at least one of the recurring units derived from an α -olefin and cyclo-olefin of 3 or more carbon atoms. It is preferably a copolymer of ethylene and an α -olefin of 3 or more carbon atoms

[0057] The component insoluble in decane at normal temperature preferably has a melt flow rate (at 190°C and a load of 2.16kg) of 0.1 to 50g/10 minutes, more preferably 0.3 to 30g/10 minutes.

[0058] The component insoluble in decane at normal temperature preferably has temperature (T_m) at the maximum peak in the endothermic curve, measured by a differential scanning calorimeter (DSC), is correlated with density (d) by the following relationship:

Tm < 400×d-250

[0059] The ethylene-based copolymer of the present invention is preferably soluble completely in decane at 64°C. [0060] It is preferable that in the ethylene-based copolymer of the present invention, when homopolypropylene having an MFR (at 190°C and a load of 2.16kg) of 40 to 60g/10 minutes is blended with ethylene copolymer so as to 75/25 of the weight ratio (homopolypropylene/ethylene-based copolymer) and then the temperature-dependence of the modulus of elasticity of the obtainable mixture is measured, the mixture have a damping factor (tanô) peak caused from the glass transition temperature of the homopolypropylene and damping factor (tanô) peak caused from the glass transition temperature of the ethylene-based copolymer, and these peaks be separated from each other. It is judged that these peaks are "separated," when the two peaks are distinctly observed, or there is a saddle between the tops of these peaks. The ethylene-based copolymer will be excellent both in impact resistance and rigidity, when it has two "separated" peaks.

[0061] The ethylene-based copolymer may be insufficient both in impact resistance and rigidity, when these peaks are not distinctly "separated" but are "fused" with each other.

[0062] The damping factor peak is confirmed by measurement of viscoelasticity, where the sample is prepared by melting and kneading the mixture of the homopolypropylene and ethylene-based copolymer which satisfy the above properties by a biaxial kneader or extruder at 200°C as cylinder temperature into the pellets and then injection-molding the pellets. The measurement of viscoelasticity is described in detail in below-mentioned EXAMPLES.

[0063] Separation of soluble-insoluble component in decane at normal temperature in the ethylene-based copolymer of the present invention can be conducted in the following manner. The sample (5g) is immersed in 200ml of decane at-room-temperature (23°C) for 5 hours with stirring, and the residue-is separated by filtration at room temperature (23°C) by a G4 glass filter into the solid phase (the insoluble component) and filtrate (the soluble component), each of which is dried. Separation of soluble-insoluble component in decane at 64°C in the ethylene-based copolymer can

be conducted in the following manner. The sample (5g) is immersed in 200ml of boiling decane for 5 hours to be dissolved, and the dissolution is cooled to 64°C. The precipitated solid component is separated by filtration by a G4 glass filter into the solid phase (the insoluble component) and filtrate (the soluble component), each of which is dried. [0064] The ethylene-based copolymer of the present invention has temperature (T_m) at the maximum peak in the endothermic curve, measured by a differential scanning calorimeter (DSC), of 40 to 90°C, preferably 45 to 90°C, more preferably 55 to 90°C.

[0065] Such an ethylene-based copolymer can be obtained by blending 1 to 70% by weight of the component [A] with 30 to 99% by weight of the component [B]. The methods for producing these components [A] and [B] are described later.

[Ethylene-based copolymer composition [II]]

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[0066] The ethylene-based copolymer composition [II] of the present invention is characterized by containing the ethylene-based copolymer [I'] which satisfies the following conditions (b-1) to (b-3).

- (b-1) Temperature (T_m) at the maximum peak in the endothermic curve of the component soluble in decane at 64°C, measured by a differential scanning calorimeter (DSC), is 40 to 90°C, preferably 50 to 90°C.
- (b-2) The copolymer [I'] contains the component soluble in decane at 64°C contains the component soluble in decane at normal temperature at 1 to 70% by weight, preferably 1 to 60%, more preferably 1 to 50%, still more preferably 1 to 30%, still more preferably 1 to 29%, wherein normal temperature means room temperature, normally 23°C.
- (b-3) The copolymer [I'] contains the component soluble in decane at normal temperature comprises (i) a recurring unit derived from ethylene and (ii) a recurring unit derived from an α -olefin of 4 or more carbon atoms, at 50 to 75% by mole for the former, preferably 55 to 75%, more preferably 60 to 72%, still more preferably 60 to 70%. The balance comprises the unit derived from an α -olefin of 4 or more carbon atoms. The component soluble in decane at normal temperature may contain the above-described cyclo-olefin, propylene or other components at 3% by mole or less as the total content.

[0067] The examples of the components soluble and insoluble in decane at normal temperature are those for the above-described ethylene-based copolymer [i].

[0068] The component soluble in decane at 64°C is the sum of the component soluble in decane at normal temperature of the above-described ethylene-based copolymer [I] and the component soluble in decane at temperature up to 64°C.

[0069] One of the examples of the component soluble in decane at 64°C but insoluble at normal temperature is the one insoluble at normal temperature for the above-described ethylene-based copolymer [I].

[0070] The ethylene-based copolymer composition [II] of the present invention preferably contains the ethylene-based copolymer [I'] at 1 to 99% by weight, more preferably 5 to 95%.

[0071] The component other than the ethylene-based copolymer [I'] is not limited for the ethylene-based copolymer composition [II], and, for example, a propylene/ α -olefin copolymer or propylene-based polymer may be used. The ethylene-based copolymer composition [II] of the present invention is preferably soluble in decane at 140°C. It may be further incorporated with an additive, e.g., inorganic filler. The concrete embodiments of the ethylene-based copolymer composition [II] are described later.

[Propylene-based copolymer composition [III]

[0072] The propylene-based copolymer composition [III] of the present invention is characterized by containing the propylene-based copolymer and satisfying the following conditions (c-1) to (c-3).

- (c-1) The copolymer contains the component soluble in decane at 64°C at 1 to 99% by weight which has a temperature (T_m) at the maximum peak in the endothermic curve, measured by a differential scanning calorimeter (DSC), of 40 to 90°C, preferably 50 to 90°C.
- (c-2) The copolymer contains the component soluble in decane at 64°C which contains the component soluble in decane at normal temperature at 1 to 70% by weight, preferably 1 to 60%, more preferably 1 to 50%, still more preferably 1 to 30%, still more preferably 1 to 29%.
- (c-3) The copolymer contains the component soluble in decane at normal temperature which comprises (i) a recurring unit derived from an α -olefin of 4 or more carbon atoms, at 50 to 75% by mole for the former, preferably 55 to 75%, more preferably 60 to 72%, still more preferably 60 to 70%. The balance comprises the unit derived from an α -olefin of 4 or more carbon atoms. The component soluble

in decane at normal temperature may contain the above-described cyclo-olefin, propylene or other compounds at 3% by mole or less as the total content.

[0073] The component-soluble in decane at normal temperature comprises (i) a recurring unit derived from ethylene and (ii) a recurring unit derived from an α -olefin of 4 or more carbon atoms, and one of the concrete examples of this component is the one for the above-described ethylene-based copolymer [I].

[0074] The component insoluble in decane at normal temperature is a polymer containing a recurring unit derived from ethylene and at least one of recurring units derived from an α -olefin and cyclo-olefin of 3 or more carbon atoms. One of the concrete examples of this component is the one for the above-described ethylene-based copolymer [I].

[0075] One of the examples of the component soluble in decane at 64°C but insoluble at normal temperature is the one insoluble at normal temperature for the above-described ethylene-based copolymer [i].

[0076] The compounds insoluble in decane at 64°C include a propylene homopolymer, and copolymer of propylene and ethylene or copolymer of propylene and an α -olefin of 4 to 20 carbon atoms. The α -olefins of 4 to 20 carbon atoms include 1-butene, 1-pentene, 4-methyl-1-pentene, 1-hexene, 1-octene, 1-decene, 1-dodecene, 1-tetradecene, 1-hexadecene, 1-octadecene, and 1-eicosene. The above α -olefin may form a random or block copolymer with propylene. Of the above compounds, the preferable ones are propylene homopolymer, crystalline propylene/ethylene block copolymer having ethylene content at 2 to 40% by mole, and crystalline propylene/ethylene random copolymer having ethylene content at 0.5 to 10% by mole.

[0077] The component insoluble in decane at 64°C has a density normally of 0.885 to 0.910g/cm³, preferably 0.890 to 0.910g/cm³, more preferably 0.895 to 0.910g/cm³.

[0078] Moreover, the component insoluble in decane at 64°C has a refractive index normally of 1.490 to 1.510, preferably 1.495 to 1.510, more preferably 1.500 to 1.510.

[0079] The component insoluble in decane at 64°C is normally preferably soluble in decane at 140°C. The propylene-based copolymer compostion may be further blended with an additive, e.g., inorganic filler.

[0080] In the present invention, the decane-insoluble component preferably contains a propylene-derived recurring unit at 90% by mole or more, more preferably 95% or more.

[0081] The copolymer composition [III] of the present invention has improved properties, in particular breaking point strength and impact resistance at low temperature, while keeping its inherent rigidity.

[0082] The composition of the present invention can be obtained by blending 1 to 99% by weight of the propylene copolymer [C] described later with a total-content of 99 to 1% by weight of the above-described polymers [A] and [B], wherein [A]/[B] ratio being kept at 1/99 to 70/30 by weight.

[Ethylene/α-olefin copolymer composition]

[0083] Next, the ethylene/α-olefin copolymer composition of the present invention is described.

[0084] The ethylene/ α -olefin copolymer composition of the present invention is composed of the ethylene/ α -olefin copolymer [A] at 1 to 70% by weight, preferably 1 to 60%, more preferably 1 to 40%, still more preferably 1 to 30%, still more preferably 3 to 29%, and ethylene-based copolymer [B] at 30 to 99% by weight, preferably 40 to 99%, more preferably 60 to 99%, still more preferably 70 to 99%, still more preferably 71 to 97%, wherein contents of the ethylene/ α -olefin copolymer [A] and ethylene-based copolymer [B] total 100% by weight.

Ethylene/α-olefin copolymer [A]

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[0085] The ethylene/ α -olefin copolymer [A] for the present invention is a copolymer of ethylene and α -olefin of 4 to 20 carbon atoms.

[0086] The examples of the α -olefins of 4 to 20 carbon atoms useful for the present invention are those for the above-described ethylene-based copolymers [I].

(A-i) Contents of the components for the ethylene/α-olefin copolymer [A]

The ethylene/ α -olefin copolymer [A] contains (a) a recurring unit derived from ethylene at 50 to 70% by mole, preferably 50 to 68%, and (b) a recurring unit derived from an α -olefin of 4 to 20 carbon atoms at 30 to 50% by mole, preferably 32 to 50%. The composition will exhibit excellent blocking properties, when each component is incorporated at the content in the above range.

The ethylene/α-olefin copolymer [A] for the present invention preferably has one of the following properties (A-ii) to (A-iv), more preferably has all-of-them:

(A-ii) Glass transition temperature and crystallinity

The ethylene/ α -olefin copolymer [A] for the present invention has a glass transition temperature of -60°C or lower, determined by a differential scanning calorimeter (DSC), preferably -65°C or lower, and a crystallinity of 1%

or less, preferably 0.5% or less. (A-iii) B value

The ethylene/ α -olefin copolymer [A] for the present invention has a B value of 0.9 to 1.5, preferably 1.0 to 1.2, given by the following general formula (1):

B value =
$$[P_{0E}]/(2[P_E][P_0])$$
 (1)

wherein, $[P_E]$ is molar fraction of the ethylene-derived recurring unit in the copolymer, $[P_0]$ is molar fraction of the α -olefin-derived recurring unit in the copolymer, and $[P_{0E}]$ is a number ratio of the ethylene/ α -olefin copolymer chains to the total dyad chains in the copolymer, determined by the ¹³C-NMR spectroscopy.

The B value is an index representing the distributions of ethylene and an α -olefin of 4 to 20 carbon atoms in the ethylene/ α -olefin copolymer [A], and can be determined by the methods proposed by, e.g., J.C. Randall (Macromolecules, 15. 353 (1982)) and J. Ray (Macromolecules, 10, 773 (1977)).

As the B value increases, the block chains in the ethylene-based copolymer become shorter, and have more uniform ethylene and α -olefin distributions, therefore copolymer rubber composition distribution becomes narrower. As the B value decreases to below 1.0, the ethylene-based copolymer composition distribution becomes broader, and possibly deteriorates its handling characteristics.

(A-iv) Intrinsic viscosity

The ethylene/ α -olefin copolymer [A] for the present invention has an intrinsic viscosity [η] of 0.1 to 10.0dl/g, determined in decalin at 135°C, preferably 1 to 8dl/g, more preferably 2 to 7dl/g.

Moreover, the ethylene/ α -olefin copolymer [A] for the present invention has the following properties (A-v) and (A-vi)

(A-v) Density

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The ethylene/ α -olefin copolymer [A] for the present invention has a density of 0.863g/cm³ or less, preferably 0.855 to 0.860g/cm³.

(A-vi) $T\alpha\beta/T\alpha\alpha$ The ethylene/ α -olefin copolymer [A] for the present invention has a $T\alpha\beta/T\alpha\alpha$ ratio of 0.5 or less, preferably 0.2 or less, more preferably less than 0.01, where $T\alpha\beta$ and $T\alpha\alpha$ are intensities determined by the ¹³C-NMR spectroscopy.

 $T\alpha\beta$ and $T\alpha\alpha$ in the ^{13}C -NMR spectroscopy are peak intensities of the two types of CH₂, different from each other in the tertiary carbon atom position as shown below, in the recurring unit derived from the α -olefin of 4 or more carbon atoms.

The intensity ratio $T\alpha\beta/T\alpha\alpha$ can be determined by the following procedure.

First. 13 C-NMR spectroscopy of the ethylene/ α -olefin copolymer [A] is measured by an analyzer (e.g., JOEL's JOEL-GX270). The measurement is conducted by using 5 % by weight of sample (copolymer) dissolved in a mixed solution of hexachlorobutadiene and d_6 -benzene (2/1 of the ratio (hexachlorobutadiene/ d_6 -benzene) by volume) at 67.8MHz and 25°C with d_6 -benzene (128ppm) as the reference. The resultant 13 C-NMR spectral pattern is analyzed by the proposals by Lindemann and Adams (Analysis Chemistry, 43, p1245 (1971)) and J.C.Randall (Review Macromolecular Chemistry Physics, C29, 201 (1989)), to find the $T\alpha\beta/T\alpha\alpha$ ratio.

The ethylene/ α -olefin copolymer [A] can be produced by copolymerization of ethylene and α -olefin of 4 to 20 carbon atoms in the presence of a metallocene catalyst.

Metallocene catalyst

[0087] The metallocene-based catalyst may be composed of a metallocene compound (a), organoaluminumoxy com-

pound (b) and/or compound (c) which reacts with the metallocene compound (a) to form the ion pair. It may be also composed of an organoaluminum compound (d) together with the compounds (a), (b) and/or (c).

[0088] Each of these components is described.

(a) Metallocene compound

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[0089] The metallocene compound for the metallocene catalyst used for the present invention is the compound of a transition metal of Group IVB in the periodic table. More concretely, it is represented by the following general formula (2):

$$ML_x$$
 (2)

wherein. M is a transition metal of Group IVB in the periodic table, "x" is its valence number, and L is a ligand. [0090] In the general formula (2), a transition metal is zirconium, titanium, hafnium or the like.

[0091] In the general formula (2), L is a ligand coordinated with the transition metal, wherein at least one ligand has a cyclopentadienyl skeleton. The ligand of cyclopentadienyl skeleton may have a substituent. The examples of the ligands of cyclopentadienyl skeleton include alkyl- or cycloalkyl-substituted clopentadienyl group, such as cyclopentadienyl, methylcyclopentadienyl, ethylcyclopentadienyl, n- or i-propylcyclopentadienyl, n-, i-, sec- or t-butylcyclopentadienyl, dimethylcyclopentadienyl, methylpropylcyclopentadienyl, methylbutylcyclopentadienyl and methylbenzylcyclopentadienyl group; and indenyl, 4,5,6,7-tetrahydroindenyl and fluorenyl group.

[0092] The above group of cyclopentadienyl skeleton may be substituted with a halogen atom or trialkylsilyl group. When the compound represented by the general formula (2) has two or more groups of cyclopentadienyl skeleton as the ligands L, these groups may be bound to each other via an alkylene group such as ethylene or propylene, substituted alkylene group such as isopropylidene or diphenylmethylene, or silylene or substituted silylene group such as dimethylsilylene, diphenylsilylene or methylphenylsilylene.

[0093] The examples of the ligand L other than that of cyclopentadienyl skeleton include a hydrocarbon group of 1 to 12 carbon atoms, alkoxy group, aryloxy group, sulfonate-containing group (e.g., -SO₃Ra, Ra is an alkyl, halogen-substituted alkyl, aryl, or halogen- or alkylsubstituted aryl), and halogen or hydrogen atom.

[0094] The examples of hydrocarbon groups of 1 to 12 carbon atoms include an alkyl, cycloalkyl, aryl and aralkyl. More concretely, they include alkyl groups, such as methyl, ethyl, n-propyl, i-propyl, n-butyl, i-butyl, sec-butyl, t-butyl, pentyl. hexyl, octyl, decyl and dodecyl; cycloalkyl groups, such as cyclopentyl and cyclohexyl; aryl groups, such as phenyl and tolyl; and aralkyl groups, such as benzyl and neophyl.

[0095] The alkoxy groups include methoxy, ethoxy and propoxy. The aryloxy groups include phenoxy. The sulfonate-containing groups (represented by -SO₃R^a) include methanesulfonate, p-toluenesulfonate, trifluoromethanesulfonate and p-chlorobenzenesulfonate. The halogen atoms include fluorine, chlorine, bromine and iodine.

[0096] More concretely, the metallocene compound represented by the general formula (2) is represented by the following general formula (3), when the transition metal is tetra-valent:

$$R^{2}_{k}R^{3}_{1}R^{4}_{m}R^{5}_{n}M$$
 (3)

wherein. M is a transition metal as is the case with the general formula (2); R^2 is a group (ligand) of cyclopentadienyl skeleton; R^3 , R^4 and R^5 are each a group (ligand) with or without cyclopentadienyl skeleton; "k" is an integer of 1 or more; and (k+l+m+n) = 4.

[0097] The examples of the metallocene compound (a) include

bis(cyclopentadienyl) zirconium dichloride, bis(methylcyclopentadienyl) zirconium dichloride, bis(ethylcyclopentadienyl) zirconium dichloride, bis(n-propylcyclopentadienyl) zirconium dichloride, bis(n-butylcyclopentadienyl) zirconium dichloride, bis(n-hexylcyclopentadienyl) zirconium dichloride, bis(methyl-n-propylcyclopentadienyl) zirconium dichloride, bis(methyl-n-butylcyclopentadienyl) zirconium dichloride,

-bis(dimethyl-n-butylcyclopentadienyl)-zirconium-dichloride,

bis(n-butylcyclopentadienyl) zirconium dibromide,

bis(n-butylcyclopentadienyl) zirconium methoxy chloride,

bis(n-butylcyclopentadienyl) zirconium ethoxy chloride, bis(n-butylcyclopentadienyl) zirconium butoxy chloride, bis(n-butylcyclopentadienyl) zirconium ethoxide, bis(n-butylcyclopentadienyl) zirconium methyl chloride, bis(n-butylcyclopentadienyl) zirconium dimethyl, bis(n-butylcyclopentadienyl) zirconium benzyl chloride, bis(n-butylcyclopentadienyl) zirconium dibenzyl, bis(n-butylcyclopentadienyl) zirconium phenyl chloride, and bis(n-butylcyclopentadienyl) zirconium hydride chloride.

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Each of these compounds may be substituted at two positions, or 1,2- and 1,3-positions.

[0098] The above metallocene compound (a) may have zirconium substituted by titanium or hafnium.

[0099] Moreover, at least two of R², R³, R⁴ and R⁵ (e.g., R² and R³) in the above general formula (3) for the metal-locene compound (a) may be groups (ligands) of cyclopentadienyl skeleton. In this case, the metallocene compound (a) may be of bridge type in which these groups are bound to each other via an alkylene, substituted alkylene, silylene or substituted silylene group, and R⁴ and R⁵ are each the ligand L other than that of cyclopentadienyl skeleton, as described in the explanation of the general formula (2).

[0100] The examples of the bridge type metallocene compound (a) include

ethylenebis(indenyl)dimethyl zirconium, ethylenebis(indenyl) zirconium dichloride, isopropylidene(cyclopentadienyl-fluorenyl) zirconium dichloride, diphenylsilylenebis(indenyl) zirconium dichloride, methylphenylsilylenebis(indenyl) zirconium dichloride, and a metallocene compound represented by the general formula (4), disclosed by Japanese Patent Laid-Open Publication No.268307/1992.

$$R^3$$
 R^5
 R^6
 $(CR^8R^9)_m$
 $(CR^8R^9)_n$

[0101] M¹ in the general formula (4) is a Group IVB metal in the periodic table, e.g., titanium, zirconium or hafnium. In the general formula (4), R¹ and R² may be the same or different. More concretely, R¹ and R² are each hydrogen atom; an alkyl group of 1 to 10, preferably 1 to 3, carbon atoms; alkoxy group of 1 to 10, preferably 1 to 3, carbon atoms; aryl group of 6 to 10, preferably 6 to 8, carbon atoms; alkenyl group of 2 to 10, preferably 2 to 4, carbon atoms; aryloxy group of 7 to 40, preferably 7 to 10, carbon atoms; alkylaryl group of 7 to 40, preferably 7 to 12, carbon atoms; arylalkenyl group of 8 to 40, preferably 8 to 12, carbon atoms; or halogen atom, preferably chlorine.

[0102] In the above general formula (4), R^3 and R^4 are each hydrogen atom; a halogen atom, preferably fluorine, chlorine or bromine; alkyl group of 1 to 10, preferably 1 to 4, carbon atoms, which may be halogenated; aryl group of 6 to 10, preferably 6 to 8, carbon atoms; or group represented by $-N(R^{10})_2$, $-SR^{10}$, $-OSi(R^{10})_3$, $-Si(R^{10})_3$ or $-P(R)_2$. R^{10} is a halogen atom, preferably chlorine; alkyl group of 1 to 10, preferably 1 to 3, carbon atoms; or aryl group of 6 to 10, preferably 6 to 8, carbon atoms. R^3 and R^4 may be the same or different, especially preferably hydrogen atom.

[0103] In the above general formula (4), each of R⁵ and R⁶ may be the same as R³ or R⁴ except hydrogen atom. R⁵ and R⁶ may the same or different, preferably the same. R⁵ and R⁶ are preferably each an alkyl group of 1 to 4 carbon atoms, or halogen-substituted alkyl group, more concretely methyl, ethyl, propyl, i-propyl, butyl, i-butyl or trifluoromethyl, of which methyl is more preferable.

[0104] In the above general formula (4), R7 is represented by

-B(R¹¹)-, A1(R¹¹), -Ge, -Sn-, -O-, -S-, =SO-, =SO₂, -N(R¹¹)-, =CO, $P(R^{11})$ - or $P(O)(R^{11})$ -.

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[0105] R¹¹, R¹² and R¹³ are each hydrogen atom; a halogen atom; alkyl group of 1 to 10, preferably 1 to 4, carbon atoms, more preferably methyl; fluoroalkyl group of 1 to 10, preferably CF₃; fluoroaryl group of 6 to 10 carbon atoms, preferably pentafluorophenyl; alkoxy group of 1 to 10, preferably 1 to 4, carbon atoms, more preferably methoxy; alkenyl group of 2 to 10, preferably 2 to 4, carbon atoms; arylalkyl group of 7 to 40, preferably 7 to 10, carbon atoms; arylalkenyl group of 8 to 40, preferably 8 to 12, carbon atoms; or alkylaryl group of 7 to 40, preferably 7 to 12, carbon atoms.

[0106] "R¹¹ and R¹²" or "R¹¹ and R¹³" may form a ring with the atom to which they are bound. R¹¹, R¹² and R¹³ may be the same or different.

[0107] M^2 is silicon, germanium or tin, preferably silicon or germanium. R^7 is $-C(R^{11})(R^{12})$ -, $Si(R^{11})(R^{12})$ -, $Ge(R^{11})(R^{12})$ -, -O-, -S-, =SO, $P(R^{11})$ - or $-P(O)(R^{11})$ -.

[0108] In the above general formula (4), R⁸ and R⁹ may be the same as R¹¹, and also may be the same as or different from each other. In the above general formula (4), "m" and "n" are each 0, 1 or 2, preferably 0 or 1; "m+n" is 0, 1 or 2, preferably 0 or 1; wherein "m" and "n" may be the same or different.

[0109] The metallocene compound (a) represented by the general formula (3) may be also rac-ethylene(2-methyl-1-indenyl)-2-zirconium dichloride or rac-dimethylsilylene(2-methyl-1-indenyl)-2-zirconium dichloride.

[0110] These metallocene compound (a) represented by the general formula (3) can be produced by a known process, e.g., the one disclosed by Japanese Patent Laid-Open Publication No.268307/1992.

[0111] The metallocene compound (a) represented by the general formula (3) may be the one represented by the following general formula (5):

[0112] In the above general formula (5), M is a Group IVB metal in the periodic table, e.g., titanium, zirconium or hafnium. In the general formula (5), R¹ and R² are each hydrogen atom; a halogen atom; hydrocarbon group of 1 to 20 carbon atoms; halogenated hydrocarbon group of 1 to 20 carbon atoms; or silicon-containing, oxygen-containing, sulfur-containing, nitrogen-containing or phosphorus-containing group. More concretely, R¹ and R² are each a halogen atom, e.g., fluorine, chlorine, bromine or iodine;

hydrocarbon group of 1 to 20 carbon atoms, such as alkyl group, e.g., methyl, ethyl, propyl, butyl, hexyl, cyclohexyl, oxtyl, nonyl, dodecyl, eicosyl, norbornyl or adamantyl; alkenyl group, e.g., vinyl, propenyl or cyclohexenyl; arylalkyl group, e.g., benzyl, phenylethyl-or-phenylpropyl; or aryl-group, e.g., phenyl, tolyl, dimethylphenyl, timethyl-phenyl, ethylphenyl, propylphenyl, biphenyl, naphthyl, methylnaphthyl, anthryl or phenanthryl;

silicon-containing group, such as mono-hydrocarbon-substituted silyl, e.g., methylsilyl or phenylsilyl; di-hydrocarbon-substituted silyl; tri-hydrocarbon-substituted silyl, e.g, trimethylsilyl, triethylsilyl, tripropylsilyl, tricyclohexylsilyl,

triphenylsilyl, dimethylphenylsilyl, methyldiphenylsilyl, tritolylsilyl or trinaphthylsilyl; ether of hydrocarbon-substituted silyl, e.g., trimethylsilyl ether; silicon-substituted alkyl, e.g., trimethylsilylmethyl; or silicon-substituted aryl, e.g., trimethylsilylphenyl;

oxygen-containing group, such as alkoxy, e.g., hydroxy, methoxy, ethoxy, propoxy or butoxy; aryloxy, e.g., phenoxy, methylphenoxy, dimethylphenoxy or naphthoxy; or arylalkoxy, e.g., phenylmethoxy or phenylethoxy;

sulfur-containing group, such as the above-described oxygen-containing group substituted with sulfur;

nitrogen-containing group, such as alkylamino group, e.g., amino, methylamino, dimethylamino, diethylamino, dipropylamino, dibutylamino or dicyclohexylamino; arylamino group, e.g., phenylamino, diphenylamino, ditolylamino, dinaphthylamino or methylphenylamino; or alkylarylamino; or

phosphorus-containing group, such as dimethylphosphino or diphenylphosphino.

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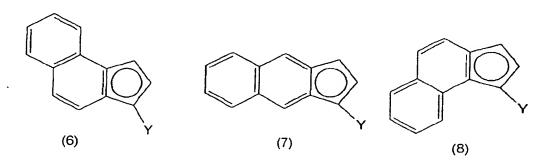
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[0113] Of these, a hydrocarbon group is preferable, especially alkyl of 1 to 3 carbon atoms, e.g., methyl ethyl or propyl. R³, R⁴, R⁵ and R⁶ are each hydrogen atom; a halogen atom; hydrocarbon group of 1 to 20 carbon atoms; halogenated hydrocarbon group of 1 to 20 carbon atoms. Each may be the same as each of R¹ and R² described above. [0114] At least one of the pairs of R³ and R⁴, R⁴ and R⁵, and R⁵ and R⁶ may be bound to each other to form a monocyclic aromatic ring, and the ligands containing the aromatic ring include those represented by the following general formulae (6) to (8):



[0115] When the metallocene compound (a) represented by the general formula (5) has two or more of R³, R⁴, R⁵ and R⁶ as hydrocarbon or halogenated hydrocarbon groups, other than those forming an aromatic ring, they may be bound to each other to form a ring. When R⁶ is a substituent other than aromatic group, it is preferably hydrogen atom. [0116] In the general formula (5), X¹ and X² are each hydrogen atom; an halogen atom; hydrocarbon group of 1 to 20 carbon atoms; halogenated hydrocarbon group of 1 to 20 carbon atoms; or oxygen- or sulfur-containing group.

[0117] The concrete examples of the halogen atom, hydrocarbon group of 1 to 20 carbon atoms, halogenated hydrocarbon group of 1 to 20 carbon atoms and oxygen-containing group are the same as the above-described R¹ or R². [0118] The examples of the sulfur-containing group include, in addition to those represented by R¹ or R², sulfonate groups, e.g., methylsulfonate, trifluoromethanesulfonate, phenylsulfonate, benzylsulfonate, methylsulfonate, p-toluenesulfonate, trimethylbenzenesulfonate, triisobutylbenzenesulfonate, p-chlorobenzenesulfonate and pentafluorobenzenesulfinate, p-toluenesulfinate, trimethylbenzenesulfinate and pentafluorobenzenesulfinate.

[0120] The concrete examples of Y include alkylene, e.g., methylene, dimethylmethylene, 1,2-ethylene, dimethylene, 1,2-ethylene, 1,3-trimethylene, 1,4-tetramethylene, 1,2-cyclohexylene and 1,4-cyclohexylene, and arylalkylene, e.g., diphenylmethylene and diphenyl-1,2-ethylene, of 1 to 20 carbon atoms;

the above described hydrocarbon group of 1 to 20 carbon atoms which is halogenated;

divalent silicon-containing group such as alkylsilylene, e.g., methylsilylene, dimethylsilylene, diethylsilylene, di (n-propyl)silylene, di(i-propyl)silylene, di(cyclohexyl)silylene, methylphenylsilylene, diphenylsilylene, di(p-tolyl)silylene, and di(p-chlorophenyl)silylene; alkyldisilylene, e.g., alkylarylsilylene, arylsilylene, tetramethyl-1,2-disilylene and tetraphenyl-1,2-disilylene; and alkylaryldisilylene and aryldisilylene;

divalent germanium-containing group which is the above-described divalent, silicon-containing group whose silicon is substituted with germanium; and

divalent tin-containing group which is the above-described divalent, silicon-containing group whose silicon is substituted with tin.

[0121] Of these, divalent silicon-, germanium and tin-containing groups are preferable, divalent silicon-containing one is more preferable, and alkylsilylene, alkylarylsilylene and arylsilylene are still more preferable.

[0122] For the compound represented by the general formula (5), a group R³ and one of R⁴ to R6 are preferably an alkyl group of 1 to 20 carbon atoms, more preferably R³ and R⁵, and R³ and R⁶. The above alkyl group is preferably secondary or tertiary, and may be substituted with a halogen atom or silicon-containing group. The examples of the halogen atom and silicon-containing group include the halogen atoms or substituents described in the R¹ or R². Moreover, each of R⁴ to R⁶ is preferably hydrogen atom, when it is not an alkyl group. The alkyl groups of 1 to 20 carbon atoms include straight-chain or cycloalkyl groups, such as methyl, ethyl, n-propyl, i-propyl, n-butyl, i-butyl, sec-butyl, tert-butyl, pentyl, hexyl, cyclohexyl, heptyl, oxtyl, nonyl, dodecyl, eicosyl, norbornyl and adamantyl; alkenyl group, such as vinyl, propenyl or cyclohexenyl; arylalkyl group, such as benzyl, phenylethyl, phenylpropyl, or tolylmethhyl. They may contain a double or triple bond. Moreover, two groups of R³, R⁴, R⁵ and R₆ may be cound to each other to form a monocyclic or polycyclic ring other than aromatic ring.

[0123] More concretely, these metallocene compounds (a) include rac-dimethylsilylene-bis(4,7-dimethyl-1-indenyl) zirconium dichloride, rac-dimethylsilylene-bis(2,4,7-trimethyl-1-indenyl) zirconium dichloride, an

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rac-dimethylsilylene-bis(2,4,7-trimethyl-1-indenyl) zirconium dichloride, and rac-dimethylsilylene-bis(2,4,6-trimethyl-1-indenyl) zirconium dichloride.

[0124] The above metallocene compound (a) may have zirconium substituted with titanium or hafnium.

[0125] The above metallocene compound (a) is normally racemic, but may be of R or S type.

[0126] Of the metallocene compounds (a) represented by the general formula (5), those which can preferably used include the ones whose R^3 group is an aryl group of 6 to 16 carbon atoms, such as phenyl, α -naphthyl, anthryl, phenanthryl, pyrenyl, acetonaphthyl, phenalenyl (perinaphthenyl), or aceanthrenyl. These aryl groups may be substituted with a halogen atom; hydrocarbon group of 1 to 20 carbon atoms or halogenated hydrocarbon group of 1 to 20 carbon atoms. Similar to R^1 described above. The especially preferable substituents are phenyl and naphthyl.

[0127] The concrete examples of these metallocene compounds include

rac-dimethylsilylene-bis(4-phenyl-1-indenyl) zirconium dichloride,

rac-dimethylsilylene-bis(2-methyl-4-phenyl-1-indenyl) zirconium dichloride,

rac-dimethylsilylene-bis(2-methyl-4- $(\alpha$ -naphthyl)-1-indenyl) zirconium dichloride,

rac-dimethylsilylene-bis(2-methyl-4-(β-naphthyl)-1-indenyl) zirconium dichloride, and

rac-dimethylsilylene-bis(2-methyl-4-(1-anthracenyl)-1-indenyl) zirconium dichloride.

[0128] The above metallocene compound may have zirconium substituted by titanium or hafnium. Moreover, it may be substituted with a transition metal compound represented by the following general formula (9):

wherein. M is a Group IVB transition metal in the periodic table, more concretely titanium, zirconium or hafnium.

[0129] In the general formula (9), R¹s may be the same or different, wherein at least one of them is an aryl of 11 to 20 carbon atoms; arylalkyl of 12 to 40 carbon atoms, arylalkenyl of 13 to 40 carbon atoms, alkylaryl of 12 to 40 carbon atoms, or silicon-containing group; or two adjacent R¹s form a monocyclic or polycyclic aromatic or aliphatic group together with the carbon atom which connects them to each other. The whole ring formed by R's has 4 to 20 of carbons including the carbon atom which is bonded with R'.

[0130] The examples of the two adjacent R1s forming a monocyclic or polycyclic aromatic or aliphatic group together with the carbon atom which connects them to each other include condensed phenyl, condensed cyclohexyl, condensed cyclopentadienyl, condensed dihyrocyclopentadienyl, condensed indenyl, condensed tetrahydroindenyl, condensed fluorenyl, condensed tetrahydrofluorenyl, and condensed octahydrofluorenyl. These groups may be substituted with

a straight-chain or cyclic alkyl, halogen atom, halogen-substituted alkyl, aryl, or silicon-, oxygen-, nitrogen- or phosphorus-containing group.

[0131] R¹ other than that forming an aryl, arylalkyl, arylalkenyl, alkylaryl group, or aromatic or aliphatic ring is hydrogen atom, a halogen atom, alkyl of 1 to 10 carbon atoms or silicon-containing group.

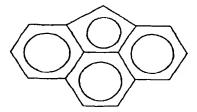
[0132] The aryl groups of 11 to 20 carbon atoms include biphenylyl, anthryl and phenanthryl; arylalkyl groups of 12 to 40 carbon atoms include phenanthrylmethyl, phenanthrylethyl and phenanthrylpropyl; arylalkenyl groups of 13 to 40 carbon atoms include biphenylphenanthryl; alkylaryl groups of 12 to 40 carbon atoms include methylphenanthryl, ethylphenanthryl and propylphenanthryl; halogen atoms include fluorine, chlorine, bromine and iodine; and alkyl groups of 1 to 10 carbon atoms include methyl, ethyl, propyl, butyl, hexyl, cyclohexyl, octyl and nonyl.

[0133] The silicon-containing groups include methylsilyl, phenylsilyl, dimethylsilyl, diethylsilyl, diphenylsilyl, trimethylsilyl, triethylsilyl, triphenylsilyl, dimethylphenylsilyl, methyldiphenylsilyl, tritolylsilyl, and trinaphthylsilyl.

[0134] The alkyl, aryl. arylalkyl, arylalkenyl and alkylaryl groups may be substituted with a halogen atom. R²s may be the same or different, and are each hydrogen atom; a halogen atom, alkyl group of 1 to 10, carbon atoms; aryl group of 6 to 20 carbon atoms; alkenyl group of 2 to 10 carbon atoms; arylalkyl group 7 to 40 carbon atoms; arylalkenyl group of 8 to 40 carbon atoms; alkylaryl group of 7 to 40 carbon atoms; or silicon-, oxygen-, nitrogen- or phosphorus-containing group.

[0135] Of the groups represented by R^2 , two adjacent groups may form a monocyclic or polycyclic aromatic or aliphatic group together with the carbon atom which connects them to each other. In this case, the ring formed by R^2 has 4 to 20 carbon atoms, including the one bound to R^2 , and R^2 other than that forming the aromatic or aliphatic ring is hydrogen atom, a halogen atom, alkyl of 1 to 10 carbon atoms or silicon-containing group.

[0136] The group in which two groups represented by R² form a monocyclic or polycyclic aromatic or aliphatic group include those having a fluorenyl group of the following structure:



The examples of the alkyl group of 1 to 10 carbon atoms and halogen atom include those described earlier.

[0138] The aryl groups of 6 to 20 carbon atoms include phenyl, biphenyl, α- and β-naphthyl, anthmethylnaphthyl, anthryl, and phenanthryl.

[0139] The arylalkyl groups of 7 to 40 carbon atoms include benzyl, phenylethyl, phenylpropyl, phenanthrylpropyl, phenanthrylpropyl.

[0140] The arylalkenyl groups of 8 to 40 carbon atoms include styryl and vinyl phenylethyl.

[0141] The alkylaryl groups of 7 to 40 carbon atoms include tolyl, dimethylphenyl, trimethylphenyl, ethylphenyl, propylphenyl, methylphenanthryl, ethylphenanthryl and propylphenanthryl.

[0142] The alkenyl groups of 2 to 10 carbon atoms include vinyl, propenyl and cyclohexenyl.

[0143] The silicon-containing groups include those described earlier.

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[0144] The oxygen-containing groups include alkoxy, e.g., hydroxy, methoxy, ethoxy, propoxy and butoxy; aryloxy, e.g., phenoxy, methylphenoxy, dimethylphenoxy and naphthoxy; and arylalkoxy, e.g., phenylmethoxy and phenylethoxy.

[0145] The sulfur-containing groups include, in addition to the oxygen-containing groups whose oxygen atom is substituted with sulfur, and sulfonate groups, such as methylsulfonate, trifluoromethanesulfonate, phenylsulfonate, benzylsulfonate, p-toluenesulfonate, trimethylbenzenesulfonate, triisobutylbenzenesulfonate, p-chlorobenzenesulfonate and pentafluorobenzenesulfonate; and sulfinate groups, such as methylsulfinate, phenylsulfinate, benzylsulfinate, p-toluenesulfinate, trimethylbenzenesulfinate and pentafluorobenzenesulfinate; and sulfinate groups.

[0146] The nitrogen-containing groups include alkylamino groups such as amino, methylamino, dimethylamino, diethylamino, dipropylamino, dibutylamino and dicyclohexylamino; arylamino groups and alkylarylamino such as phenylamino, diphenylamino, ditolylamino, dinaphthylamino or methylphenylamino.

[0147] The phosphorus-containing groups include dimethylphosphino and diphenylphosphino.

[0148] Of these, R² is preferably hydrogen atom and an alkyl group, more preferably hydrogen atom and an alkyl group of 1 to 3 carbon atoms, such as methyl ethyl or propyl.

[0149] As the fluorenyl group having an R² as the substituent, 2,7-dialkyl-fluorenyl group is cited as the preferable one, and the examples of alkylgrow constituted 2,7-dialkyl are an alkyl of 1 to 5 carbon atoms.

[0150] R¹ and R² may be the same or different. R³ and R⁴ may be also the same or different. More concretely, R³ and R⁴ are each hydrogen atom; a halogen atom; alkyl group of 1 to 10 carbon atoms; aryl group of 6 to 20 carbon atoms; alkenyl group of 2 to 10 carbon atoms; arylalkyl group of 7 to 40 carbon atoms; arylalkenyl group of 8 to 40 carbon atoms; alkylaryl group of 7 to 40 carbon atoms; and silicon-containing, oxygen-containing, sulfur-containing, nitrogen-containing or phosphorus-containing groups, as is the case described earlier.

[0151] Of these, at least one of R3 and R4 is an alkyl of 1 to 3 carbon atoms.

[0152] X¹ and X² may be also the same or different, and are each hydrogen atom; an halogen atom; hydrocarbon group of 1 to 20 carbon atoms; halogenated hydrocarbon group of 1 to 20 carbon atoms; oxygen-, sulfur- or nitrogen-containing group; or a conjugated diene residue formed by X¹ and X². More concretely, the halogen atoms, and oxygen-, sulfur- and nitrogen-containing groups are similar to the atoms and groups as described earlier.

[0153] The hydrocarbon groups of 1 to 20 carbon atoms include alkyl groups, e.g., methyl. ethyl, propyl, butyl, hexyl, cyclohexyl, oxtyl, nonyl, dodecyl, eicosyl, norbornyl and adamantyl; alkenyl groups, e.g., vinyl, propenyl and cyclohexenyl; arylalkyl groups, e.g., benzyl, phenylethyl and phenylpropyl; and aryl group, e.g., phenyl, tolyl, dimethylphenyl, trimethylphenyl, ethylphenyl, propylphenyl, α- or β-naphthyl, methylnaphthyl, anthryl, phenanthryl, benzylphenyl, pyrenyl, acenaphthyl, phenalenyl, aceanthrenyl, tetrahydronaphthyl, indanyl and biphenylenyl. The halogenated hydrocarbon groups of 1 to 20 carbon atoms include the above-described hydrocarbon groups of 1 to 20 carbon atoms which are substituted with a halogen atom.

[0154] The examples of the conjugated diene residue formed by X^1 and X^2 include η^4 -1,4-diphenyl-1,3-butadiene, η^4 -1,4-dibenzyl-1,3-butadiene, η^4 -1-phenyl-1,3-pentadiene, η^4 -3-methyl-1,3-pentadiene, η^4 -1,4-bis (trimethylsilyl)-1,3-pentadiene, 2.3-dimethylbutadiene, η^4 -2,4-hexadiene, and isoprene.

[0155] The preferable examples of the conjugated diene residue formed by X^1 and X^2 include those of 1,3-butadiene, 2,4-hexadiene, 1-phonyl-1,3-pentadiene and 1,4-diphenylbutadiene, which may be substituted with a hydrocarbon group of 1 to 10 carbon atoms.

[0156] Of these as X^1 and X^2 , a halogen atom, hydrocarbon group of 1 to 20 carbon atoms or sulfur-containing group is preferable. Y is a divalent hydrocarbon group of 1 to 20 carbon atoms; divalent hydrocarbon group of 1 to 20 carbon atoms; divalent silicon-containing group; divalent germanium-containing group; divalent tin-containing group; or -O-, -CO-, -S-, -SO-, -SO₂-, -NR⁵-, P(R⁵)-, P(O)(R⁵)-, BR⁵- or -AIR⁵- (R⁵ is hydrogen atom, a halogen atom, hydrocarbon group of 1 to 20 carbon atoms, or halogenated hydrocarbon group of 1 to 20 carbon atoms).

[0157] The concrete examples of Y include divalent hydrocarbon group having 1 to 20 carbon atoms such as alkylene, e.g., methylene, dimethylene, 1,2-ethylene, dimethyl-1,2-ethylene, 1,3-trimethylene, 1,4-tetramethylene, 1,2-cyclohexylene and 1,4-cyclohexylene; and arylalkylene, e.g., diphenylmethylene and diphenyl-1,2-ethylene;

the above described hydrocarbon group of 1 to 20 carbon atoms which is halogenated;

divalent silicon-containing group such as alkylsilylene, e.g., methylsilylene, dimethylsilylene, diethylsilylene, di (n-propyl)silylene, di(i-propyl)silylene, di(cyclohexyl)silylene, methylphenylsilylene, diphenylsilylene, di(p-tolyl)silylene, and di(p-chlorophenyl)silylene; alkylarylsilylene; arylsilylene; alkyldisilylene, e.g., tetramethyl-1,2-disilylene and tetraphenyl-1,2-disilylene; and alkylaryldisilylene and aryldisilylene;

divalent germanium-containing group which is the above-described divalent, silicon-containing group whose silicon is substituted with germanium; and

divalent tin-containing group which is the above-described divalent silicon-containing group whose silicon is substituted with tin.

[0158] Of these divalent groups, the preferable ones have the shortest bonding section is composed of 1 or 2 atoms for -Y-, represented by the general formula (I). R⁵ is a halogen atom, hydrocarbon group of 1 to 20 carbon atoms or halogenated hydrocarbon group of 1 to 20 carbon atoms, similarly to the one described earlier.

[0159] Of these groups, -Y- is preferably a divalent hydrocarbon group of 1 to 5 carbon atoms, and divalent silicon-and germanium-containing groups, more preferably divalent silicon-containing group, still more preferably alkylsilylene, alkylarylsilylene and arylsilylene.

[0160] The above-described zirconium compound may have zirconium substituted with titanium or hafnium.

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[0161] Of the above-described zirconium compounds, dimethylsilylene(2,7-dimethyl-4,5-(2-methyl-benzo)-1-indenyl)(2.7-di-t-butyl-9-fluorenyl) zirconium dichloride, for example, has the following structure.
 [0162] Another example is dimethylsilylene(2,6-dimethyl-4,5-(1-methyl-benzo)-1-indenyl)(2,7-di-t-butyl-9-fluorenyl) zirconium dichloride, whose structure is shown below.

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[0163] These catalyst can be synthesized by the method proposed by the inventors of this invention, disclosed by Japanese Patent File No.187563/1996 (Japanese Patent Laid-Open Publication No.235313/1997).

[0164] A metallocene compound (a) represented by the following general formula (10) may be also used for the

[U164] A metallocene compound (a) represented by the following general formula (10) may be also used for the present invention:

$$L^{a}MX_{2}$$
 (10) (10)

wherein. M is a Group IV or lanthanide metal in the periodic table; L^a is a derivative of nonlocalized π bonding group, which imparts a constrained geometric shape to the active site of the metal M; and X's are each hydrogen, a halogen, hydrocarbon group containing 20 atoms or less of carbon, silicon or germanium, or silyt or germyl group.

[0165] Of the compounds represented by the general formula (10), the compounds represented by the following general formula (11) are preferable:

$$C_{p}$$
 $(X)_{2}$ (11)

wherein. M is titanium, zirconium or hafnium; X is the same as that for the general formula (10); Cp is a substituted cyclopentadienyl group with a substituent Z and bound to M through the π bonding; Z is oxygen, sulfur, boron or a

Group IVA element in the periodic table, e.g.,..silicon, germanium or tin; and Y is a logand containing nitrogen, phosphorus, oxygen or sulfur, wherein Z and Y may together form a condensed ring.

[0166] The compounds represented by the general formula (11) include (dimethyl) (t-butylamide) (tetramethyl- η^5 -cyclopentadienyl)silane)-titanium dichloride, and ((t-butylamide)(tetramethyl- η^5 -cyclopentadienyl)-1,2-ethanediyl) titanium dichloride. The above metallocene compounds may have titanium substituted by zirconium or hafnium. Of the metallocene compounds (a) represented by the general formula (10) or (11), those preferably used ones include zirconocene compounds having zirconium as the metal atom at the center and a ligand containing at least 2 cyclopentadienyl skeletons.

[0167] The metallocene compounds (a) represented by the general formula (4) or (5) preferably have titanium as the metal atom at the center.

[0168] These metallocene compounds may be used either individually or in combination for the present invention. The metallocene compound (a) may be diluted with a hydrocarbon or halogenated hydrocarbon for use. It may be also used after being brought into contact with a carrier.

[0169] The carriers useful for the present invention are of inorganic or organic compounds in the form of solid, granules or fine particles, of 10 to 300 μm in size, preferably 20 to 200 μm. Of these inorganic carriers, porous oxides are preferable. The concrete examples include SiO₂, Al₂O₃, MgO, ZrO₂, TiO₂, B₂O₃, CaO, ZnO, BaO, ThO₂ and a mixture thereof, e.g., SiO₂-MgO, SiO₂-Al₂O₃, SiO₂-TiO₂, SiO₂-V₂O₅, SiO₂-Cr₂O₃, SiO₂-TiO₂-MgO. Of these, more preferable ones have at least one of SiO₂ and Al₂O₃ as the main component.

[0170] The above inorganic oxide may be incorporated with a small quantity of carbonate, sulfate, nitrate or oxide, e.g., Na₂CO₃, K₂CO₃, CaCO₃, MgCO₃, Na₂SO₄, Al₂(SO₄)₃, BaSO₄, KNO₄, Mg(NO₃)₂, Al₂(NO₃)₃, Na₂O, K₂O, or Li₂O. [0171] These carriers suitably used for the present invention have a specific surface area of 50 to 1000m²/g, preferably 100 to 700m²/g, and pore volume of 0.3 to 2.5m²/g, although having varying properties depending on type and method by which they are produced. They may be fired, as required, at 100 to 1000°C, preferably 150 to 700°C before being used.

[0172] The carrier for the present invention may be also of a solid organic compound, granules or fine particles, having a particle size of 10 to $300\mu m$. The concrete examples include polymers or copolymers produced from an α -olefin of 2 to 14 carbon atoms as the main component, e.g., ethylene, propylene, 1-butene and 4-methyl-1-pentene, or polymers or copolymers produced from vinyl cyclohexene or styrene as the main component.

Organoaluminumoxy compound (b)

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[0173] Next. the organoaluminumoxy compound (b) used for producing a metallocene-based catalyst is described. [0174] The organoaluminumoxy compound (b) useful for the present invention may be a known aluminoxane or benzene-insoluble organoalumnumoxy compound (b).

[0175] More concretely, the known aluminoxanes are represented by the following general formula (12) or (13):

$$R_2$$
—AIH—(OAI)_m— $OAIR_2$ (12)
 R (13)

wherein. R is a hydrocarbon group, e.g., methyl, ethyl, propyl or butyl, preferably methyl or ethyl, more preferably methyl; and "m" is an integer of 2 or more, preferably 5 to 40.

[0176] The aluminoxane represented by the general formula (12) or (13) may be formed by a mixture of alkyloxy aluminum units represented by the general formulae (OAI(R¹) and (OAI(R²), wherein R¹ and R² are similar to those represented by R, but different from each other.

[0177] The organoaluminumoxy-compound (b)-useful for the present invention may be contained with a small quantity—of organometallic compound of a metal other than aluminum. The above-described aluminoxanes can be produced by, e.g., one of the following method:

- (1) To suspend in which a compound containing adsorbed water or a salt containing bonding water, e.g., magnesium chloride hydrate, copper sulfate hydrate, aluminum sulfate hydrate, nickel sulfate hydrate or cerium sulfate hydrate, are suspended in a hydrocarbon solvent, organoaluminum compound is added, thereby the organoaluminum compound is reacted with the compound containing absorbed water or the salt containing bonding water, e.g., trialkyl aluminum, and then the product is recovered as the solution of hydrocarbon.
- (2) An organoaluminum compound, e.g., trialkyl aluminum, is directly reacted with water, ice or steam in a solvent, e.g., benzene, toluene, ethyl ether or tetrahydrofuran, and the product is recovered as the solution of hydrocarbon.
- (3) An organoaluminum compound, e.g., trialkyl aluminum, is reacted with an organotin oxide, e.g., dimethyl tin oxide or dibutyl tin oxide, in a solvent, e.g., decane, benzene or toluene.

[0178] The alumoxane may contain a small quantity of an organometallic component. The aluminoxane may be redissolved in a solvent, after the solution recovered is distilled to remove the solvent and unreacted organoaluminum compound.

[0179] The concrete examples of the organoaluminum compounds useful for production of the aluminoxane include trialkyl aluminum, such as trimethyl aluminum, triethyl aluminum, tripropyl aluminum, triisopropyl aluminum, tri-n-butyl aluminum, triisobutyl aluminum, tri-sec-butyl aluminum, tri-tert-butyl aluminum, tripentyl aluminum, trihexyl aluminum, trioctyl aluminum, and tridecyl aluminum; tricycloalkyl aluminum, such as tricyclohexyl aluminum and tricycloactyl aluminum;

dialkyl aluminum halide, such as dimethyl aluminum chloride, diethyl aluminum chloride, diethyl aluminum chloride; mide, and diisobutyl aluminum chloride;

dialkyl aluminum hydride, such as diethyl aluminum hydride and diisobuty aluminum hydride; dialkyl aluminum alkoxide, such as dimethyl aluminum methoxide and diethyl aluminum methoxide; dialkyl aluminum aryloxide, such as diethyl aluminum phenoxide.

[0180] Of these, trialkyl aluminum and tricycloalkylaluminum are particularly preferable.

[0181] The isoprenyl aluminum represented by the following general formula (14) is also the organoaluminum compound which can be used for the present invention:

$$(i-C_4H_9)_xAI_y(C_5H_{10})_z$$
 (14)

wherein. "x", "y" and "z" are each a positive numeral, and $z \ge 2x$.

[0182] The above organoaluminum compounds may be used either individually or in combination. The solvents useful for preparation of the aluminoxane include aromatic hydrocarbons, such as benzene, toluene, xylene, cumene and cymene; aliphatic hydrocarbons, such as pentane, hexane, heptane, octane, decane, dodecane, hexadecane and octadecane; alicyclic hydrocarbons, such as cyclopentane, cyclohexane, cyclooctane and methylcyclopentane; petroleum fractions, such as gasoline, kerosene and gas oil; and halides of the above aromatic, aliphatic and alicyclic hydrocarbons, in particular chlorides and bromides. The other useful compounds include ethers, e.g., ethyl ether and tetrahydrofuran. Of these solvents, aromatic hydrocarbons are more preferable.

[0183] The above benzene-insoluble organoaluminumoxy compound is insoluble or hardly soluble in benzene, having the aluminum component soluble in benzene at 60°C is 10% or less as atomic aluminum, preferably 5% or less, more preferably 2% or less.

[0184] Solubility of the organoaluminumoxy compound in benzene is determined by the following procedure. The organoaluminumoxy compound in a quantity corresponding to 100mg of atomic aluminum is suspended in 100ml of benzene, stirred at 60°C for 6 hours for mixing, and filtered under heating at 60°C by a jacketed filter of G-5 glass. The separated cake on the filter is washed with 50ml of benzene kept at 60°C 4 times, and quantity (x mmol) of the atomic aluminum present in the totaled filtrate is measured (x %).

Ionized ionic compound (c)

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[0185] The examples of the ionized ionic compounds (c) (sometimes referred to as ionic ionized or ionic compounds) useful for the present invention include Lewis acids, ionic compounds, borane compounds and carborane compounds. [0186] The Lewis acids useful for the present invention include Mg-, Al- and B-containing ones, of which B-containing ones are more preferable. The B-containing Lewis acids include those represented by the general formula BR₃ (R is a phenyl group which may be substituted with fluorine, methyl or trifluoromethyl, or fluorine). More concretely, they include trifluoroboron, triphenylboron, tris(4-fluorophenyl) boron, tris(3.5-difluorophenyl) boron, tris(4-fluoromethylphenyl) boron, tris(pentafluorophenyl) boron, tris(pentafluorophenyl) boron, tris(o-tolyl) boron, and tris(3,5-dimethylphenyl) boron. [0187] The ionic compound is a salt of cationic and anionic compounds, the anion reacting with the above-described

metallocene compound (c) to cationize it and to form the ion pair, thereby working to stabilize the transition metal cationic species. The examples of the anion include organoboron, organoarsenic and organoaluminum compound anions. They are preferably relatively bulky and work to stabilize the transition metal cationic species. The examples of the cation include metallic, organometallic, carbonium (carbenium) oxonium, sulfonium, phosphonium and ammonium cation. In more detail, they include triphenylcarbenium, tributylammonium, N,N-dimethylammonium and ferrocenium cation.

[0188] The concrete examples of these ionic compounds include trialkyl-substituted ammonium salts, such as N,N-dialkylanilinium, dialkylammonium and triarylphosphonium salt;

trialkyl-substituted ammonium salts, such as triethylammonium tetra(phenyl) boron, tripropylammonium tetra(phenyl) boron, tri(n-butyl) ammonium tetra(phenyl) boron;

dialkyl ammonium salts, such as di(1-propyl)ammoniumtetra(pentafluorophenyl) boron and dicyclohexylammoniumtetra(ohenyl) boron; and

other boron-containing ionic compounds, e.g., triphenylcarbeniumtetrakis(pentafluorophenyl) borate, N,N-dimethylaniliniumtetrakis(pentafluorophenyl) borate and ferroceniumtetra(pentafluorophenyl) borate;

borane compounds such as decaborane (14), bis[tri(n-butyl)ammonium]nonaborate, metallic borane anion salts, e.g., nickel acid salts (III) of bis[tri(n-butyl)ammonium]bis(dodecahydridedodecaborate)decaborate; carborane compounds such as 4-carbanonaborane (14), 1,3-dicarbanonaborane (13), metallic carborane anion salts, e.g., nickel acid salts (IV) of bis[tri(n-butyl)ammonium]bis(undecahydride-7-carbaundecaborate)decaborate.

[0189] The above ionized ionic compounds (c) may be used either individually or in combination. The organoaluminumoxy compound (b) and ionized ionic compound (c) may be used in such a way that the above-described carrier is impregnated therewith.

[0190] The organoaluminum compound (d) may be used together with the organoaluminumoxy compound (b) or ionized ionic compound (c) for production of the metallocene-based catalyst.

Organoaluminum compound (d)

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[0191] The organoaluminum compounds (d) which may be used as required for production of the metallocene-based catalyst include the organoaluminum compounds represented by the following general formula (15):

$$R^{1}_{n}AIX_{3-n}$$
 (15)

wherein. R¹ is a hydrocarbon group of 1 to 12 carbon atoms; X is a halogen or hydrogen; and "n" is 1 to 3. In the general formula (15), R¹ is a hydrocarbon group of 1 to 12 carbon atoms, e.g., an alkyl, cycloalkyl or aryl group, more concretely, methyl, ethyl, n-propyl, isopropyl, isobutyl, pentyl, hexyl, octyl, cyclopentyl, cyclohexyl, phenyl or tolyl.

[0192] The concrete examples of these organoaluminum compounds (d) include the following compounds:

trialkylaluminum, such as trimethyl aluminum, triethyl aluminum, triisopropy aluminum, triisobutyl aluminum, trioctyl aluminum. and tri-2-ethylhexyl aluminum;

alkenyl aluminum, such as isoprenyl aluminum;

dialkyl alminum halide, such as dimethyl aluminum chloride, diethyl aluminum chloride, diisopropyl aluminum chloride, diisobutyl aluminum chloride, and dimethyl aluminum bromide;

alkyl aluminum sesquihalide. e.g., methyl aluminum sesquichloride, ethyl aluminum sesquichloride, isopropyl aluminum sesquichloride, butyl aluminum sesquichloride, and ethyl aluminum sesquibromide;

alkyl aluminum dihalide, e.g., methyl aluminum dichloride, ethyl aluminum dichloride, isopropyl aluminum dichloride, and ethyl aluminum dibromide; and

alkyl aluminum hydride, e.g., diethyl aluminum hydride and diisobutyl aluminum hydride.

[0193] The organoaluminum compound represented by the following general formula (15) can be used as the organoaluminum compounds (d) for production of the metallocene-based catalyst:

$$R^{1}_{n}AIX_{3-n}$$
 (16)

The concrete examples of the organoaluminum compounds include:

- (1) compounds represented by the general formula $R^1_nAl(OR^2)_{3-n}$, e.g., dimethyl aluminum methoxide, diethyl aluminum methoxide and diisobutyl aluminum methoxide;
- (2) compounds represented by the general formula $R^1_nAl(OSiR^2_3)_{3-n}$, e.g., $Et_2Al(OSiMe_3)$, (iso-Bu)₂Al(OSiMe₃); and (iso-Bu)₂Al(OSiEt₃);
- (3) compounds represented by the general formula $R^1_nAl(OAlR^4_2)_{3-n}$, e.g., $Et_2AlOAlEt_3$ and (iso-Bu)₂AlOAl(iso-Bu)₂;
- (4) compounds represented by the general formula $R^1_nAl(NR^5_2)_{3-n}$, e.g., Me_2AlNEt_2 , $Et_2AlNHMe$, $Me_2AlNHEt$, $Et_2AlN(SiMe_3)_2$;
- (5) compounds represented by the general formula $R_{n}^{1}Al(SiR_{2}^{6})_{3-n}$, e.g., (iso-Bu)₂AlSiMe₃; and
- (6) compounds represented by the general formula $R_n^1Al(N(R^7)AlR_2)_{3-n}$, e.g., $Et_2AlN(Me)AlEt_2$ and (iso-Bu)₂AlN (Et)Al(iso-Bu)₂.
- [0194] Of the compounds represented by the general formula (15) or (16), the more preferable ones are those represented by the general formula R₃Al, R_nAl(OR²)_{3-n} or R_nAl(OAIR₂)_{3-n}, still more preferably those with R of isoalkyl and n=2.
 - [0195] Ethylene and an α -olefin of 4 to 20 carbon atoms are copolymerized normally in the liquid phase in the presence of a calalyst composed of the above-described metallocene catalyst composed of the metallocene compound (a), an organoaluminumoxy compound (b) and/or ionized ionic compound (c), and, optionally organoaluminum compound (d). A hydrocarbon solvent is normally used for the above copolymerization process, but the α -olefin per se may be used as the solvent.
 - [0196] The copolymerization process is effected either batchwise, semicontinuously or continuously. For the batch process, the components of the catalyst are controlled at the following contents.
 - [0197] When the metallocene catalyst composed of the metallocene compound (a), an organoaluminumoxy compound (b) and/or ionized ionic compound (c) is used, content of the of the metallocene compound (a) is normally set at 0.00005 to 0.1mmol/l (polymerization volume) in the polymerization system, preferably 0.0001 to 0.05mmol/l. The organoaluminumoxy compound (b) is supplied at 1 to 10000 as the atomic ratio of the aluminum to the transition metal in the metallocene compound (a) in the polymerization system (Al/transition metal atomic ratio), preferably 10 to 5000.
 - [0198] The ionized ionic compound (c) is supplied at 0.5 to 20 as the molar ratio of the ionized ionic compound (c) to the metallocene compound (a) in the polymerization system (ionized ionic compound (c)/metallocene compound (a) molar ratio). preferably 1 to 10.
 - [0199] The organoaluminum compound (d), when used, is supplied to control its content at around 0 to 5mmol/l (polymerization volume), preferably around 0 to 2mmol/l.
 - **[0200]** The copolymerization process is effected normally at -20 to 150°C, preferably 0 to 120°C, more preferably 0 to 100°C, and higher than the atmospheric pressure but 7.8MPa (80kgf/cm²G) or lower, preferably higher than the atmospheric pressure but 4.9MPa (50kgf/cm²G) or lower.
 - [0201] Ethylene and α -olefin are supplied to the polymerization system at rates to obtain the ethylene/ α -olefin copolymer [A] of the above-mentioned specified composition. A molecular weight adjusting agent, such as hydrogen, may be used for the copolymerization process.
 - [0202] Copolymerization of ethylene and α -olefin under the above conditions normally gives a polymerization effluent containing an ethylene-based copolymer. It is treated by the normal procedure to produce the ethylene-based copolymer of the present invention.
 - [0203] The ethylene-based copolymer of the present invention is used as a resin modifier, incorporated in the ethylene/ α -olefin copolymer, propylene-based polymer or the like.

Ethylene-based copolymer [B]

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- [0204] The ethylene-based copolymer [B] used for the present invention comprises ethylene and at least one compound selected from the group consisting of α -olefin and cyclo-olefin-based compound of 3 to 20 carbon atoms.
- [0205] The concrete examples of the α -olefins useful for the present invention include propene, 1-butene, 1-pentene, 1-hexene, 3-methyl-1-butene, 3-methyl-1-pentene, 3-ethyl-1-pentene, 4-methyl-1-pentene, 4-methyl-1-hexene, 4,4-dimethyl-1-pentene, 4-ethyl-1-hexene-1-octene, 3-ethyl-1-hexene, 1-octene, 1-decene, 1-dodecene, 1-tetra-decene, 1-hexadecene, 1-octadecene, and 1-eicosene, of which propene, 1-butene, 1-hexene, 4-methyl-1-pentene and 1-octene are more preferable.
- [0206] The cyclo-olefins useful for the present invention include those represented by the general formula (1-1) or (1-2).
- [0207] A polyene may be used, in addition to ethylene, an α -olefin and/or cyclo-olefin-based compound of 3 to 20,

within limits not harmful to the object of the present invention.

[0208] The examples of the polyenes useful for the present invention include cyclic polyenes, such as 5-ethylidene-2-norbornene, 5-vinyl-2-norbornene, 5-isopropenyl-2-norbornene, dicyclopentadiene, and nonconjugated triene and tetraene, of which 5-ethylidene-2-norbornene and dicyclopentadiene are preferable. These cyclic polyene-based compounds may be used either individually or in combination.

[0209] The ethylene-based copolymer [B] used for the present invention preferably comprises 80 to 93% by mole of an ethylene-derived recurring unit and 1 to 20% by mole of at least one compound selected from the group consisting of α -olefin or cyclo-olefin-based compound of 3 to 20 carbon atoms.

10 (B-i) Density

[0210] The ethylene-based copolymer [B] used for the present invention has a density of 0.870 to 0.895g/cm³, preferably 0.870 to 0.890g/cm³, more preferably 0.875 to 0.890g/cm³. When incorporated in a resin, e.g., polypropylene, as a resin modifier, the ethylene-based copolymer [B] having a density in the above range will give the composition well-balanced in rigidity and impact resistance. The modifier having a density beyond the above range may give the composition of deteriorated impact strength and excessive hardness.

[0211] The ethylene-based copolymer [B] used for the present invention preferably satisfies the following properties (B-ii) and (B-iii):

20 (B-ii) Melt flow rate

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[0212] The ethylene-based copolymer [B] used for the present invention has a melt flow rate (190°C and a load of 2.16kg) of 0.1 to 50g/10 minutes, preferably 0.3 to 30g/10 minutes.

(B-iii) Relationship between T_m and density (d)

[0213] The ethylene-based copolymer [B] used for the present invention satisfies the following relationship between T_m and density (d), wherein T_m is temperature at the maximum peak in the endothermic curve, measured by a differential scanning calorimeter (DSC):

 $T_m < 400 \times d-250$,

preferably

 $T_m < 400 \times d-297$

more preferably

 $T_m < 500 \times d-344$

and still more preferably

 $T_m < 500 \times d-391$.

[0214] Moreover, the ethylene-based copolymer [B] used for the present invention preferably satisfies, in addition to the above properties, the following relationship between melt tension (MT) at room temperature and melt flow rate (MFR, 190°C and a load of 2.16kg):

 $\mathrm{MT} \leq 2.20 \times \mathrm{MFR}^{\text{-}0.84}$

[0215] Moreover, the ethylene-based copolymer [B] used for the present invention preferably satisfies the following relationship between fraction of the n-decane-soluble component at room temperature (W, % by weight) and density (d):

(a) at an MFR of 10g/10 minutes or less:

 $W < 80 \times exp(-100(d - 0.88)) + 0.1$

more preferably

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 $W < 60 \times exp(-100(d - 0.88)) + 0.1$

still more preferably

 $W < 40 \times exp(-100(d - 0.88)) + 0.1$.

More preferably, it also satisfies the following relationship: (b) at an MFR higher than 10g/10 minutes:

 $W < 80 \times (MFR - 9)^{0.26} \times exp(-100(d - 0.88)) + 0.1.$

[0216] The ethylene-based copolymer [B] can be produced by copolymerization of ethylene and at least one compound selected from the group consisting of α -olefin and cyclo-olefin-based compound of 3 to 20 carbon atoms in the presence of the metallocene catalyst. The metallocene catalysts useful for the present invention include:

bis(cyclopentadienyl) zirconium dichloride, bis(methylcyclopentadienyl) zirconium dichloride, bis(ethylcyclopentadienyl) zirconium dichloride, bis(n-propylcyclopentadienyl) zirconium dichloride, bis(n-butylcyclopentadienyl) zirconium dichloride, bis(n-hexylcyclopentadienyl) zirconium dichloride, bis(methyl-n-propylcyclopentadienyl) zirconium dichloride, bis(methyl-n-butylcyclopentadienyl) zirconium dichloride,

bis(methyl-n-butylcyclopentadienyl) zirconium dichloride, bis(dimethyl-n-butylcyclopentadienyl) zirconium dichloride.

bis(n-butylcyclopentadienyl) zirconium dibromide.

bis(n-butylcyclopentadienyl) zirconium methoxy chloride,

bis(n-butylcyclopentadienyl) zirconium ethoxy chloride,

bis(n-butylcyclopentadienyl) zirconium butoxy chloride,

bis(n-butylcyclopentadienyl) zirconium ethoxide,

bis(n-butylcyclopentadienyl) zirconium methyl chloride,

bis(n-butylcyclopentadienyl) zirconium dimethyl,

bis(n-butylcyclopentadienyl) zirconium benzyl chloride,

bis(n-butylcyclopentadienyl) zirconium dibenzyl,

bis(n-butylcyclopentadienyl) zirconium phenyl chloride, and

bis(n-butylcyclopentadienyl) zirconium hydride chloride.

Each of these compounds may be substituted at two positions, or 1,2- and 1,3-positions. The above metallocene compound (a) may have zirconium substituted by titanium or hafnium.

Ethylene/α-olefin copolymer composition

[0217] For the ethylene/α-olefin copolymer composition

[A] for the present invention, it is preferable that ratio of density (d_b) of the ethylene-based copolymer [B] to density ($d_{\bar{a}}$) of the composition-[A]-($d_{\bar{b}}/d_a$) be 1.05 or less, preferably 1:04 or less. Its lower limit is preferably

1.01, more preferably 1.02.

[0218] The ethylene/ α -olefin copolymer composition [A] having the (d_b/d_a) ratio in the above range has excellent

transparency and breaking elongation, and exhibits excellent properties as a modifier.

[0219] The ethylene/ α -olefin copolymer composition of the present invention is useful as a modifier for thermoplastic resins to improve their impact resistance and rigidity. These thermoplastic resins include polyolefin, polyamide, polyester, polystyrene, polyvinyl chloride and polyvinyl alcohol. When used for modification of a thermoplastic resin containing a polar group, the ethylene/ α -olefin copolymer composition of the present invention may be graft-modified with an unsaturated carboxylic acid.

[0220] The ethylene/ α -olefin copolymer composition of the present invention, comprising a specific ethylene/ α -olefin copolymer and specific ethylene-based copolymer, can improve balances between rigidity and breaking point strength/impact resistance of the thermoplastic resin for which it is used as a modifier, these effects being especially noted when it is used for polypropylene. It also shows good workability, facilitating the modification process without causing blocking or other troubles.

[0221] When used as a modifier for a resin. e.g., polypropylene, the ethylene/ α -olefin copolymer composition of the present invention is preferably handled by an extruder or the like which continuously kneads and discharges the resin composition. It is preferable to effect the kneading process at softening temperature or melting point of the resin being discharged or higher but 400°C or lower.

[0222] It is recommended, when the ethylene/ α -olefin copolymer composition of the present invention is used as a modifier for a resin, e.g., polypropylene, that it be pelletized by extrusion or injection molding and blended in the molten resin to be modified.

[0223] The ethylene/ α -olefin copolymer composition of the present invention, when used as a modifier for a resin, e.g., polypropylene, can improve properties of the resin, in particular impact resistance at low temperature and tensile elongation strength, while keeping its inherent rigidity.

[0224] The method of the present invention for resin modification is characterized by blending the pellets of the ethylene/ α -olefin copolymer composition and the resin to be modified in molten state. This method can modify a thermoplastic resin to have well-balanced rigidity and breaking point strength/impact resistance without causing blocking or other troubles. As such, it is an excellent thermoplastic resin modification method in productivity and workability.

[Propylene-based polymer composition]

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[0225] The propylene-based polymer composition of the present invention comprises a propylene-based polymer, and the ethylene/ α -olefin copolymer [A] and ethylene-based copolymer [B], namely the propylene-based polymer and ethylene/ α -olefin copolymer composition [AB].

First propylene-based polymer composition (1)

35 [0226] The first propylene-based polymer composition (1) of the present invention is comprised of the propylene-based polymer [C-1], ethylene/α-olefin copolymer [A] and ethylene-based copolymer [B], namely the propylene polymer [C-1] and ethylene/α-olefin copolymer composition [AB].

[0227] It contains the propylene-based polymer [C-1] at 99 to 1% by weight, preferably 98 to 60%. more preferably 95 to 65%, and ethylene/ α -olefin copolymer [A] and ethylene-based copolymer [B] at 1 to 99% by weight (total content of [A] and [B]), preferably 2 to 40%, more preferably 5 to 35%.

[0228] The content ratio of the ethylene/ α -olefin copolymer [A] to the ethylene-based copolymer [B], ([A]/[B] content ratio), is 1/99 to 70/30, 3/97 to 70/30 as the ratio of one of the preferred embodiments, still more preferably 3/97 to 29/71, as is the case with the above-described ethylene/ α -olefin copolymer.

[0229] It is preferable that the propylene-based copolymer have a damping factor ($\tan\delta$) peak caused from the glass transition temperature of the propylene-based polymer [C-1] and another damping factor ($\tan\delta$) peak caused from the glass transition temperature of the ethylene/ α -olefin copolymer composition (composed of the ethylene/ α -olefin copolymer [A] and ethylene-based copolymer [B]), when the temperature-dependence of the modulus of elasticity is measured at intervals of 3°C, and that these peaks be separated from each other. It is judged that these peaks are "separated," when the two peaks are distinctly observed, or there is a saddle between the tops of these peaks. The propylene-based polymer composition is excellent both in impact resistance and rigidity, when it has two "separated" peaks.

[0230] The propylene-based polymer composition may be insufficient both in impact resistance and rigidity, when it has these peaks that are not distinctly "separated" but are "fused" with each other.

[0231] The propylene-based polymer [C-1] preferably has an MFR (190°C and a load of 2.16kg) of 0.1g/10 minutes or more, more preferably 0.5 to 200g/10 minutes, still more preferably 1 to 200g/10 minutes.

[0232] Use of propylene-based-polymer-[C-1]-can-give well-balanced rigidity, tensile breaking point elongation and/or impact-resistant characteristics, and also excellent in fluidity to a propylene-based polymer composition.

[0233] The propylene-based polymer [C-1] is a propylene homopolymer, or a copolymer of propylene and ethylene or an α -olefin of 4 to 20 carbon atoms. The α -olefins of 4 to 20 carbon atoms include ethylene, 1-butene, 1-pentene,

4-methyl-1-pentene, 1-hexene, 1-octane, 1-decene, 1-dodecene, 1-tetradecene, 1-hexadecene, 1-octadecene, and 1-eicosene. These α -olefins may be used either individually or in combination. Each α -olefin may form a random or block copolymer with propylene.

[0234] The preferable polymers for the present invention are propylene homopolymer, crystalline propylene/ethylene block copolymer containing ethylene at 2 to 40% by mole and crystalline propylene/ethylene random copolymer containing ethylene at 0.5 to 10% by mole are preferable.

[0235] By blending the propylene-based polymer [C-1] having an MFR of 4g/10 minutes or more, the ethylene/ α -olefin copolymer composition can be improved properties, especially breaking point strength and impact resistance, while keeping the inherent rigidity of the polymer itself.

[0236] The propylene-based polymer [C-1] preferably has a density normally of 0.885 to 0.910g/cm³, more preferably 0.890 to 0.910g/cm³, still more preferably 0.895 to 0.910g/cm³.

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[0237] Moreover, the propylene-based polymer [C-1] has a refractive index normally of 1.490 to 1.510, preferably 1.495 to 1.510, more preferably 1.500 to 1.510.

[0238] The propylene-based polymer [C-1] having these properties can be produced by various processes, for example, in the presence of a catalyst composed of solid titanium and an organometallic compound as the catalyst components, high-activity titanium catalyst composed of an electron donor in addition to the above two components, catalyst composed of a metallocene compound and aluminoxane, or a combination thereof. In case that the propylene-based polymer [C-1] is a block copolymer, different catalysts may be used for different steps in the multi-stage polymerization process.

[0239] The method for producing the olefin-based polymer composition (1) of the present invention is not limited. For example, the ethylene/ α -olefin copolymer [A], ethylene-based copolymer [B] and propylene-based polymer [C-1] may be kneaded altogether, or the propylene-based polymer [C-1] and the ethylene/ α -olefin copolymer composition may be kneaded by a known method using an internal mixer, e.g., Banbury mixer, kneader or intermix. The latter method is more preferable for the present invention because of its workability.

[0240] A styrene-based thermoplastic elastomer may be added for production of the propylene-based polymer composition (1) within limits not harmful to the object of the present invention. The styrene-based thermoplastic elastomers useful for the present invention include a block copolymer of a styrene compound and conjugated diene. The styrenes useful for the present invention include alkyl styrenes, e.g., styrene, α-methyl styrene and p-methyl styrene, p-methoxy styrene, vinyl naphthalene, and a combination thereof. Of these, styrene is more preferable.

[0241] The conjugated diene compounds useful for the present invention include butadiene, isoprene, piperylene, methylpentadiene, phenylbutadiene, 3,4-dimethyl-1,3-hexadiene, 4,5-diethyl-1,3-octadiene, and a combination thereof, of which butadiene and isoprene are more preferable.

[0242] The concrete examples of these styrene-based themoplastic elastomers include styrene/butadiene diblock copolymer, styrene/butadiene/styrene triblock copolymer, styrene/isoprene diblock copolymer, styrene/isoprene/styrene triblock copolymer, hydrogenated styrene/butadiene/styrene triblock copolymer, hydrogenated styrene/isoprene diblock copolymer, and hydrogenated styrene/isoprene/styrene triblock copolymer.

[0243] The styrene-based thermoplastic elastomer for the present invention preferably comprises the recurring unit derived from the styrene-based compound and that derived from the conjugated diene compound in a weight ratio of 10/90 to 65/35, preferably 20/80 to 50/50.

[0244] The molecular structure of the styrene-based thermoplastic elastomer may be of straight-chain, branched, radiating, or a combination thereof. The propylene-based polymer composition of the present invention may comprise, in addition to the above-described ethylene/α-olefin copolymer composition and propylene-based polymer [C-1], one or more additives, e.g., nucleating agent, antioxidant, hydrochloric acid absorber, softening agent, light stabilizer, ultraviolet absorber, lubricant, antiaging agent, processing aid, heat stabilizer, weathering stabilizer, antistatic agent, flame retardant, pigment, dye, dispersant, copper inhibitor, neutralizer, foaming agent, plasticizer, antifoaming agent, crosslinking agent, flow improver (e.g., peroxide), weld strength improver, and antifogging agent. Moreover, it may be further blended with a known inorganic filler, e.g., talc or glass fibers, preferably at 1 to 40 parts by weight per 100 parts by weight of the propylene-based polymer composition.

[0245] The propylene-based polymer composition (1) of the present invention can be formed into various shapes, e.g., film, sheet and pipe, by a forming method, e.g., extrusion, injection molding, inflation or calendering. The formed article is excellent in recovery from the strained conditions.

[0246] The propylene-based polymer composition of the present invention, comprising a specific ethylene/α-olefin copolymer and specific ethylene-based copolymer, has well-balanced resin rigidity and breaking point strength/impact resistance.

Second propylene-based polymer composition (2)

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[0247] The second propylene-based polymer composition (2) of the present invention is composed of the propylene-based polymer [C-2], ethylene/ α -olefin copolymer [A], ethylene-based copolymer [B] and an inorganic filler [D]. In other words, it comprises the propylene-based polymer, ethylene/ α -olefin copolymer composition and an inorganic filler, and hence falls into the category of the propylene-based polymer composition (1).

[0248] In particular, the ethylene/ α -olefin copolymer [A] for the second propylene-based polymer composition (2) of the present invention has the following characteristics, and is specifically referred to as [A-1]:

[A-1] Copolymer of ethylene and α -olefin of 4 to 20 carbon atoms (ethylene/ α -olefin copolymer) comprises (i) 50 to 70% by mole of (a) a recurring unit derived from the ethylene and 30 to 50% by mole of (b) a recurring unit derived from the α -olefin of 4 to 20 carbon atoms, and (ii) has an intrinsic viscosity [η] of 0.1 to 10.0dl/g, determined in decalin at 135°C.

The ethylene-based copolymer [B] for the second propylene-based polymer composition (2) of the present invention has the following characteristics, and is specifically referred to as [B-1]:

[B-1]: Copolymer of (a) ethylene and (b) at least one compound selected from the group consisting of an acolefin and cyclo-olefin of 3 to 20 carbon atoms, has (i) a density of 0.870 to 0.895g/cm³ and (ii) a melt flow rate (190°C and a load of 2.16kg) of 0.1 to 50g/10 minutes.

[0249] The ethylene/ α -olefin-based copolymer composition [AB] is composed of the ethylene/ α -olefin-based copolymer [A] at 1 to 50% by weight, preferably 3 to 40%, more preferably 3 to 30%, still more preferably 3 to 29%, and the copolymer [B] at 50 to 99% by weight, preferably 60 to 97%, more preferably 70 to 97%, still more preferably 71 to 97%, based on the whole composition of [A] and [B].

[0250] Combination of the cthylene/ α -olefin-based copolymer composition [AB] and inorganic filler [D] brings about the favorable effects of improving rigidity and tensile breaking point elongation while keeping balances between them. [0251] It is adequate that the propylene-based polymer composition of the present invention be composed of the propylene-based polymer [C-2] at 20 to 93% by weight, preferably 30 to 90%, more preferably 40 to 80%; ethylene/ α -olefin copolymer composition [AB] at 6 to 79%, preferably 10 to 70%, more preferably 20 to 60%; and inorganic filler [D] at 1 to 25%, preferably 5 to 20%, wherein the sum of the propylene-based polymer [C-2], ethylene/ α -olefin copolymer composition [AB] and inorganic filler [D] is 100% by weight.

[0252] The propylene-based polymer composition of the present invention has sufficiently improved properties, in particular impact resistance at low temperature and tensile elongation strength, when it contains the ethylene/ α -olefin copolymer composition [AB] at higher than the above-described lower limit, and is less likely to lose the excellent original properties of the propylene-based polymer [C-2], when it contains [AB] at lower than the above-described upper limit. The propylene-based polymer composition has sufficiently improved properties, in particular rigidity and heat resistance, when it contains the inorganic filler [D] at higher than the above-described lower limit, and is less likely to lose the characteristics, e.g., moldability and outer appearances, when it contains [D] at lower than the above-described upper limit.

[0253] The ethylene-based copolymer [A-1] preferably used for the present invention (A-i) comprises (a) a recurring unit derived from ethylene in the ethylene-based copolymer [A] at 50 to 70% by mole, preferably 50 to 68%, and (b) a recurring unit derived from an α -olefin of 4 to 20 carbon atoms at 30 to 50% by mole, preferably 32 to 50%. The composition has excellent blocking characteristics with each recurring unit present in the above range.

[0254] Moreover, the ethylene-based copolymer [A-1] used for the present invention preferably satisfies the above-described characteristics (A-ii), (A-iii), (A-v) and (A-vi).

[0255] The ethylene-based copolymer [B-1] used for the present invention has (B-i) a density of 0.870 to 0.895g/cm³, preferably 0.870 to 0.890g/cm³; more preferably 0.875 to 0.890g/cm³. When incorporated in a resin, e.g., polypropylene, as a resin modifier, the ethylene-based copolymer [B-1] having a density in the above range will give the composition well-balanced in rigidity and impact resistance.

[0256] The ethylene-based copolymer [B-1] used for the present invention has a melt flow rate (at 190°C and a load of 2.16kg) of 0.1 to 50g/10 minutes, preferably 0.3 to 30g/10 minutes.

[0257] Moreover, the ethylene-based copolymer [B-1] used for the present invention preferably satisfies the above-described property (B-iii).

[0258] In the present invention, the specific propylene-based polymer [C-2] having the following characteristics is used. The propylene-based polymer [C-2] may be a homopolypropylene, or propylene-based block or random copolymer so long as it satisfies the following characteristics, but is preferably homopolypropylene or propylene-based block-copolymer.

[0259] The propylene-based block copolymer is preferably composed of the highly crystalline polypropylene component (crystalline component) and ethylene/propylene copolymer rubber component soluble in n-decane at room

temperature (23°C) (rubber component).

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[0260] The propylene-based copolymer [C-2] has a melt flow rate (determined in accordance with ASTM D1238 at 230°C and a load of 2.16kg) of 0.1 to 400g/10 minutes, preferably 0.1 to 200g/10 minutes.

[0261] The propylene-based copolymer [C-2] having a melt flow rate in the above range can give the propylene-based copolymer composition which is excellent in fluidity and can be formed into a large-sized shape.

[0262] Moreover, the propylene-based copolymer [C-2] used for the present invention contains the component soluble in n-decane at room temperature (23°C) (rubber component) at 0.01 to 30% by weight, preferably 0.1 to 30%, more preferably 0.1 to 20%. The component can sufficiently exhibit the effect of improving impact resistance when contained at 0.01% by weight or more, and the effect of improving rigidity when contained at 30% by weight or less.

[0263] The component soluble in n-decane at room temperature has an intrinsic viscosity [η] of 0.2 to 10dl/g, determined in decalin at 135°C, preferably 0.2 to 8dl/g.

[0264] Moreover, the component soluble in n-decane at room temperature preferably contains the ethylene-derived recurring unit at 30 to 50% by mole, preferably 30 to 45% by mole.

[0265] The component soluble in n-decane at room temperature is the rubber component for the propylene-based copolymer [C-2], and is preferably an isotactic polypropylene or ethylene/propylene copolymer.

[0266] The component soluble in n-decane at room temperature for the propylene-based copolymer [C-2] may be contained a unit derived from a polymerizable compound other than ethylene or propylene, within the limits not harmful to the object of the present invention.

[0267] These polymerizable compounds useful for the present invention, other than ethylene or propylene, include α -olefins, such as 1-butene, 1-pentene, 1-hexene, 1-heptane, 1-octene, 1-decene, 1-decene, 1-hexadecene, 4-methyl-1-pentene; vinyl compounds, such as vinyl cyclopentane, vinyl cyclohexane and vinyl norbornane; vinyl ester, such as vinyl acetate; and unsaturated organic acids, such as maleic anhydride or its derivative.

[0268] Content of the component soluble in n-decane at room temperature for the propylene-based copolymer [C-2] can be determined by the following procedure:

[0269] 5g of the sample (propylene polymer) is immersed in 200ml of boiling decane for 5 hours to be dissolved therein and cooled to room temperature, and the separated solid phase is filtered by a G4 glass filter and dried, to determine the content from the solid phase weight by the back calculation.

[0270] The propylene-based copolymer [C-2] contains the component insoluble in n-decane at room temperature at 99.99 to 70% by weight, preferably 99.9 to 70%, more preferably 99.9 to 80%.

[0271] The component insoluble in n-decane at room temperature is the highly crystalline polypropylene component (isotactic polypropylene) in the propylene-based polymer. More concretely, it has a pentad isotacticity I₅ of 0.95 or more, determined by the ¹³C-NMR spectroscopy, preferably 0.97 or more.

[0272] Pentad isotacticity I_5 is an isotactic fraction of the polypropylene molecular chain, determined by the ¹³C-NMR (nuclear magnetic resonance) spectroscopy proposed by A. Zambelli et al, Macromolecules <u>6</u>, 925 (1973). It is a fraction of the propylene monomer unit expressed in the pentad unit, in which a series of 5 propylene monomer units are bound to each other in an isotactic manner.

[0273] The compounds to which the NMR peaks are relevant are determined by the method described in Macromolecules 8, 687 (1975). The ¹³C-NMR spectroscopy is measured by a Fourier transformed NMR analyzer (500MHz for measurement of the hydrogen nuclear), and the signal detection limit of the ¹³C-NMR spectroscopy can be improved to 0.001 by integrating 20,000 measurements at 125MHz.

[0274] The propylene polymer with the component insoluble in n-decane at room temperature having an I_5 level in the above range can give the composition excellent in rigidity.

[0275] In propylene-based polymer [C-2], the polymer which contains homopolymers or copolymers of 3-methyl-1-butene, 3,3-dimethyl-1-butene, 3-methyl-1-hexene, 3,5,5-trimethyl-1-hexene, vinyl cyclopentene, and vinyl norbornene as a prepolymer prepolymerized, has high crystallized rate.

[0276] The propylene-based polymer [C-2] for the present invention can be produced by various processes, e.g., in the presence of a catalyst of high stereoregularity.

[0277] More concretely, it can be produced in the presence of a catalyst composed of solid titanium, an organometallic compound and, optionally, electron donor as the catalyst components.

[0278] The concrete examples of the solid titanium catalyst component useful for the present invention include the solid titanium catalyst in which titanium trichloride or its composition is supported by a carrier having a specific surface area of 100m²/g or more, and the solid titanium catalyst in which magnesium, a halogen, electron donor (preferably aromatic carboxylate ester or alkyl-containing ether) and titanium as the essential components are supported by a carrier having a specific surface area of 100m²/g or more. Of these, the latter being more preferable.

[0279] The organometallic-compounds-useful for the present invention as the catalyst component include organoaluminum compounds, e.g., trialkyl aluminum, dialkyl aluminum hydride, alkyl aluminum sesqui-halide, and alkyl aluminum dihalide. An adequate organoaluminum compound may be selected, depending on type of the titanium component used.

[0280] The electron donors useful for the present invention include organic compounds containing nitrogen, phosphorus, silicon or boron, preferably the esters or ethers containing one of the above elements.

[0281] These catalysts may be activated by an adequate method, such as co-grinding and the above olefin may be prepolymerized onto the catalysts.

[0282] The concrete examples of the inorganic filler [D] useful for the present invention include:

powdered fillers, including finely powdered natural silicic acids and silicates, e.g., talc, kaolinite, fired clay, pyrophyllite, sericite and wollastonite; carbonates, e.g., precipitating calcium carbonate, limestone powder and magnesium carbonate; hydroxides, e.g., aluminum hydroxide and magnesium hydroxide; oxides, e.g., zinc oxide, zinc white and magnesium oxide; and synthetic silicic acids and silicates, e.g., hydrated calcium silicate, hydrated aluminum silicate, hydrated silicic acid and anhydrous silicic acid; flaky fillers, e.g., mica;

fibrous fillers, e.g., basic magnesium sulfate whisker, calcium titanate whisker, aluminum borate whisker, sepiolite, processed mineral fiber (PMF), xonotlite, potassium titanate and ellestadite; and balloon fillers, e.g., glass balloons and fry ash balloons.

[0283] Of these, talc is more preferable for the present invention, especially finely powdered talc having an average particle size of 0.01 to $10\mu m$.

[0284] The average particle size may be determined by the liquid sedimentation method.

[0285] The inorganic filler for the present invention, in particular talc, may be untreated or surface-treated beforehand. When surface treated, it may be treated chemically with a silane coupling agent, higher fatty acid, polyethylene glycol or the like, or physically. The blend of the surface-treated inorganic filler to the composition gives high weld strength, coating characteristics and moldability to automobile interiors and exteriors and gasoline tanks.

[0286] The above inorganic fillers may be used either individually or in combination.

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[0287] The composition of the present invention may be incorporated with, in addition to the inorganic filler, an organic filler, e.g., high styrenes, lignin or recycled rubber.

[0288] Combination of the specific ethylene/ α -olefin-based copolymer composition [AB] and inorganic filler [D] makes the second propylene-based copolymer composition (2) of the present invention suitable as a modifier for propylene, improving its rigidity and tensile breaking point elongation while keeping balances between them. By the blend of the specific ethylene/ α -olefin copolymer composition [AB] and inorganic filler [D] to the above-described propylene-based polymer [C-2], the propylene-based polymer composition having the desired characteristics is obtained.

[0289] In the modification of the resin, it is preferable to use an extruder or the like which continuously kneads and discharges the resin composition. It is also preferable to effect the kneading process at softening temperature of the resin being discharged or higher but 400°C or lower.

[0290] It is adequate that the propylene-based polymer composition of the present invention is composed of the propylene-based polymer [C-2] at 20 to 93% by weight, preferably 30 to 90%, more preferably 40 to 80%; ethylene/ α -olefin copolymer composition [AB] at 6 to 79%, preferably 10 to 70%, more preferably 20 to 60%; and inorganic filler [D] at 1 to 25%, preferably 5 to 20%, wherein the sum of the propylene-based polymer [C-2], ethylene/ α -olefin copolymer composition [AB] and inorganic filler [D] is 100% by weight.

[0291] The propylene-based polymer composition of the present invention has sufficiently improved properties, in particular impact resistance at low temperature and tensile elongation strength, when it contains the ethylene/ α -olefin copolymer composition [AB] as the modifier at higher than the above-described lower limit, and is less likely to lose the excellent properties the propylene-based polymer [C-2] inherently have, when it contains [AB] at lower than the above-described upper limit.

[0292] The propylene-based polymer composition has sufficiently improved properties, in particular rigidity and heat resistance, when it contains the inorganic filler [D] at higher than the above-described lower limit, and is less likely to lose the characteristics, e.g., moldability and outer appearances, when it contains [D] at lower than the above-described upper limit.

[0293] It is preferable that the obtained propylene-based copolymer composition have a damping factor (tanδ) peak caused from the glass transition temperature of the propylene-based polymer [C-2] and that caused from the glass transition temperature of the ethylene/α-olefin copolymer composition, when the temperature-dependence of the modulus of elasticity is measured at intervals of 3°C, and that these peaks be separated from each other. It is judged that these peaks be "separated," when the two peaks are distinctly observed, or there is a saddle between the tops of these peaks. The propylene-based copolymer which has two "separated" peaks is excellent both in impact resistance and rigidity.

[0294] The propylene-based polymer may be insufficient both in impact resistance and rigidity, when these peaks are not distinctly "separated" but are "fused" with each other.

[0295] The propylene-based polymer composition of the present invention may be incorporated with one or more

additives within limits not harmful to the object of the present invention. The additives useful for the present invention include a nucleating agent, antioxidant, hydrochloric acid absorber, heat stabilizer, light stabilizer, ultraviolet absorber, lubricant. antistatic agent, flame retardant, pigment, dye, dispersant, copper inhibitor, neutralizer, foaming agent, plasticizer. antifoaming agent, crosslinking agent, flow improver (e.g., peroxide), and weld strength improver.

[0296] As antioxidants, phenol-based antioxidant, sulfur-based antioxidant and phosphorus-based antioxidant and the like can be used.

[0297] More concretely, the phenol-based antioxidants include phenols, e.g., 2,6-di-tert-butyl-p-cresol, stearyl (3.3-dimethyl-4-hydroxybenzyl)thioglycolate, stearyl-β-(4-hydroxy-3,5-di-tert-butylphenol)propionate, distearyl-3,5-di-tert-butyl-4-hydroxybenzylsulfonate, 2,4,6-tris(3',5'-di-tert-butyl-4'-hydroxybenzylthio)-1,3,5-triazine, stearyl(4-hydroxy-3-methyl-5-tert-butylbenzyl)malonate, 2,2'-methylenebis(4-methyl-6-tert-butylphenol), 4,4'-methylenebis (2.6-di-tert-butylphenol), 2,2'-methylenebis[6-(1-methylcyclohexyl)p-cresol], bis[3,5-bis[4-hydroxy-3-tert-butylphenyl) butylic acid[glycol ester, 4,4'-butylidenebis(6-tert-butyl-m-cresol), 1,1,3-tris(2-methyl-4-hydroxy-5-tert-butylphenyl) butane. bis[2-tert-butyl-4-methyl-6-(2-hydroxy-3-tert-butyl-5-methylbenzyl)phenyl] terephthalate, 1,3,5-tris(2,6-dimethyl-3-hydroxy-4-tert-butyl)benzylsocyanurate, 1,3,5-tris(3,5-di-tert-butyl-4-hydroxybenzyl)-2,4,6-trimethylbenzene, tetrakis[methylene-3-(3,5-di-tert-butyl-4-hydroxyphenyl)propionate]methane, 1,3,5-tris(3,5-di-tert-butyl-4-hydroxybenzyl)isocyanurate, 2-octylthio-4,6-di(4-hydroxy-3.5-di-tert-butyl)phenoxy-1,3,5-triazine, and 4,4'-thiobis(6-tert-butyl-m-cresol); and

polyvalent phenol/carbonate oligoesters, e.g., that of 4,4'-butylidenebis(2-tert-butyl-5-methylphenol) having a degree of polymerization of, e.g., 2 to 10.

[0298] The sulfur-based antioxidants useful for the present invention include dialkyl thiodipropionates, e.g., dilauryl, dimyristyl or distearyl thiodipropionates; and esters of alkyl thiopropionates (e.g., butyl, oxtyl, lauryl or stearyl thiopropionates) and polyhydric alcohols (e.g., glycerin, trimethylol ethane, trimethylol propane, pentaerythritol and trishydroxyethylisocyanurate).

[0299] The phosphorus-based antioxidants useful for the present invention include trioctyl phosphite, trilauryl phosphite. tridecyl phosphite, octyl-diphenyl phosphite, tris(2,4-di-tert-butylphenyl) phosphite, triphenyl phosphite, trisbutoxyethyl phosphite. tris(nonylphenyl) phosphite, distearylpentaerythritol diphosphite, tetra(tridecyl)-1,1,3-tris(2-methyl-5-tert-butyl-4-hydroxyphenyl) butane diphosphite, tetraalkyl (a mixed alkyl of 12 to 15 carbon atoms)-4,4'-isopropylidenediphenyl diphosphite, tetra(tridecyl)-4,4'-butylidenebis(3-methyl-6-tert-butylphenol) diphosphite, tris(3,5-di-tert-butyl-4-hydroxyphenyl) phosphite, tris(a mixed mono- and di-nonylphenyl) phosphite, hydrogenated 4,4'-isopropylidenediphenol polyphosphite. bis(octylphenyl) -bis[4,4'-butylidenebis(3-methyl-6-tert-butylphenol-1,6-hexanediol diphosphite. phenyl-4.4'-isopropylidenediphenol-pentaerythritol diphosphite, bis(2,4-di-tert-butylphenyl) pentaerythritol diphosphite. bis(2.6-di-tert-butyl-4-methylphenyl) pentaerythritol diphosphite, tris[4,4'-isopropylidenebis(2-tert-butylphenol)] phosphite. phenyl-diisodecyl phosphite, di(nonylphenyl)pentaerythritol diphosphite, tris[1,3-di-stearoy-loxyisopropyl] phosphite. 4.4'-isopropylidenebis(2-tert-butylphenol) di(nonylphenyl) diphosphite, 9,10-di-hydro-9-oxa-9-oxa-10-phsphaphenanthrene-10-oxide, and tetrakis(2,4-di-tert-butylphenyl)-4,4'-biphenylene diphosphonite.

[0300] The other antioxidants useful for the present invention include 6-hydroxychromane derivatives, e.g., α -, β -, γ - and δ -tocopherols and a mixture thereof, 2,5-dimethyl-substituted 2-(4-methyl-pentan-3-enyl)-6-hydroxychromane, 2.5.8-trimethyl-substituted 2-(4-methyl-pentan-3-enyl)-6-hydroxychromane and 2,5,7,8-tetramethyl-substituted 2-(4-methyl-pentan-3-enyl)-6-hydroxychromane,

2.2.7-trimethyl-5-tert-butyl-6-hydroxychromane, 2,2,5-trimethyl-7-tert-butyl-6-hydroxychromane, 2,2,5-trimethyl-6-tert-butyl-6-hydroxychromane, and 2,2-dimethyl-5-tert-butyl-6-hydroxychromane.

[0301] The compounds represented by the general formula

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$$M_XAI_Y(OH)_{2x+3y-2z}$$
 (A)_z.aH₂O:

wherein. M is Mg. Ca or Zn; A is an anion other then hydroxyl; "x,"-"y" and "z" are each a positive number; and "a" is 0 or a positive number.

may be used as the hydrochloric acid absorbers for the present invention. These compounds include

Mg₁₀Al₂(OH)₂₂CO₂.4H₂O,

 $Mg_6Al_2(OH)_{16}HPO_4.4H_2O$,

Ca₆Al₂(OH)₁₆CO₃.4H₂O,

 $Zn_6Al_2(OH)_{16}CO_3.4H_2O$,

Zn₆Al₂(OH)₁₆SO₄.4H₂O,

Mg₆Al₂(OH)₁₆SO₃.4H₂O,

and

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 $Mg_6Al_2(OH)_{112}CO_3.3H_2O.$

[0302] The light stabilizers useful for the present invention include hydroxybenzophenones, e.g., 2-hydroxy-4-methoxybenzophenone, 2-hydroxy-4-n-octoxybenzophenone-2,2'-di-hydroxy-4-methoxybenzophenone and 2,4-dihydroxybenzophenone; benzotriazoles, e.g., 2-(2'-hydroxy-3'-tert-butyl-5'-methylphenyl)-5-chlorobenzotriazole, 2-(2'-hydroxy-3'-5'-methylphenyl)-5-chlorobenzotriazole and 2-(2'-hydroxy-3',5'-di-tert-amylphenyl)-5-chlorobenzotriazole;

benzoates. e.g., phenyl salycylate, p-tert-butylphenyl salycylate, 2,4-di-tert-butylphenyl-3.5-di-tert-butyl-4-hydroxybenzoate, and hexadecyl-3,5-di-tert-butyl-4-hydroxybenzoate;

nickel compounds, e.g., Ni salts of 2,2'-thiobis(4-tert-octylphenol), [2,2'-thiobis(4-tert-octylphenolate)-n-butylam:ne Ni and Ni salt of (3,5-di-tert-butyl-4-hydroxybenzyl) phosphonic acid monoethyl ester; substituted acrylonitriles, e.g. α -cyano- β -methyl- β -(p-methoxyphenyl) methyl acrylate;

oxalic acid dianilides, e.g., N'-2-ethylphenyl-N-ethoxy-5-tert-butylphenyl oxalic acid diamide and N-2-ethylphenyl-N'-2-ethoxyphenyl oxalic acid diamide; and

hindered amine compounds, e.g., bis(2,2,6,6-tetramethyl-4-piperidine) sebacate, poly[{6-(1,1,3,3-tetramethyl-butylimino}hexamethylene], condensate of dimethyl succinate and 2-(4-hydroxy-2,2.6,6-tetramethyl-1-piperidyl) ethanol

[0303] The lubricants useful for the present invention include aliphatic hydrocarbons, e.g., paraffin wax, polyethylene wax and polypropylene wax; higher fatty acids, e.g., capric acid, lauric acid, myristic acid, palmitic acid, margaric acid, stearic acid, arachidic acid and behenic acid or salts thereof (e.g., lithium, calcium, sodium, magnesium or potassium salt):

aliphatic alcohols, e.g., palmitic alcohol, cetyl alcohol and stearyl alcohol;

aliphatic amides, e.g., caproic acid amide, caprilic acid amide, capric acid amide, lauric acid amide, myristic acid amide. palmitic acid amide and stearic acid amide; esters of an aliphatic compound and alcohol; and fluorine compounds. e.g.. fluoroalkyl carboxylic acid or metallic salt thereof and metallic salts of fluoroalkyl sulfonates.

[0304] Each of the above additives may be blended at 0.0001 to 10 parts by weight per 100 parts by weight of the propylene-based polymer composition.

[0305] The propylene-based polymer composition of the present invention can form moldings further improved in properties, e.g., property balances, durability, coating characteristics, printing characteristics, resistance to scratching and moldability, by blending one or more of the above additives.

[0306] The propylene-based polymer composition of the present invention may be blended with a nucleating agent, as described above.

[0307] Various known nucleating agents may be used for the present invention without any restriction. Of these, the nucleating agents of aromatic-phosphate-ester-salts and dibenzylidene sorbitol described below-are-preferable.

wherein, R1 is oxygen, sulfur or a hydrocarbon group of 1 to 10 carbon atoms; R2 and R3 are each hydrogen or a hydrocarbon group of 1 to 10 carbon atoms, where R2 and R3 may be the same or different, and R2 and another R2, R3 and another R3 or R2 and R3 may be bound to each other to form a ring; M is a mono- to tri-valent metallic atom; and "n" is an integer of 1 to 3.

[0308] More concretely, they include:

sodium-2,2'-methylene-bis(4,6-di-tert-butylphenyl) phosphate, 20 sodium-2,2'-ethylidene-bis(4,6-di-tert-butylphenyl) phosphate, lithium-2,2'-methylene-bis(4,6-di-tert-butylphenyl) phosphate, lithium-2,2'-ethylidene-bis(4,6-di-tert-butylphenyl) phosphate, sodium-2,2'-cthylidene-bis(4-i-propyl-6-tert-butylphenyl) phosphate, 25 lithium-2,2'-methylene-bis(4-methyl-6-tert-butylphenyl) phosphate, lithium-2,2'-methylene-bis(4-ethyl-6-tert-butylphenyl) phosphate, calcium-bis[2,2'-thiobis(4-methyl-6-tert-butylphenyl) phosphate, calcium-bis[2,2'-thiobis(4-ethyl-6-tert-butylphenyl) phosphate, calcium-bis[2,2'-thiobis(4,6-di-tert-butylphenyl) phosphate, 30 magnesium-bis[2,2'-thiobis(4,6-di-tert-butylphenyl) phosphate. magnesium-bis[2,2'-thiobis(4-tert-octylphenyl) phosphate, sodium-2,2'-butylidene-bis(4,6-di-methylphenyl) phosphate, sodium-2,2'-butylidene-bis(4,6-di-tert-butylphenyl) phosphate, sodium-2,2'-tert-octylmethylene-bis(4.6-di-methylphenyl) phosphate, 35 sodium-2,2'-tert-octylmethylene-bis(4.6-di-tert-butylphenyl) phosphate, calcium-bis-(2,2'-methylene-bis(4,6-di-tert-butylphenyl) phosphate, magnesium-bis-[2,2'-methylene-bis(4.6-di-tert-butylphenyl) phosphate], barium-bis-[2,2'-methylene-bis(4,6-di-tert-butylphenyl) phosphate], sodium-2,2'-methylene-bis(4-methyl-6-tert-butylphenyl) phosphate, 40 sodium-2,2'-methylene-bis(4-ethyl-6-tert-butylphenyl) phosphate, sodium (4,4'-dimethyl-5,6'-di-tert-butyl-2,2'-biphenyl) phosphate, calcium-bis[(4,4'-dimethyl-6,6'-di-tert-butyl-2,2'-biphenyl) phosphate]. sodium-2,2'-ethylidene-bis[(4-m-butyl-6-tert-butylphenyl) phosphate, sodium-2,2'-methylene-bis(4,6-di-methylphenyl) phosphate, sodium-2,2'-methylene-bis(4,6-di-ethylphenyl) phosphate, potassium-2,2'-ethylidene-bis(4,6-di-tert-butylphenyl) 45 phosphate, calcium-bis[2,2'-ethylidene-bis(4,6-di-tert-butylphenyl) phosphate], magnesium-bis[2,2'-ethylidene-bis(4,6-di-tert-butylphenyl) phosphate], barium-bis[2,2'-ethylidene-bis(4,6-di-tert-butylphenyl) phosphate], 50 aluminum-tris[2,2'-methylene-bis(4,6-di-tert-butylphenyl) phosphate], aluminum-tris[2,2'-ethylidene-bis(4,6-di-tert-butylphenyl) phosphate], and a combination thereof. Of these, sodium-2,2'-methylene-bis(4,6-di-tert-butylphenyl) phosphate is more preferable.

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$$\left[\left(R^4 - O \right) - O \right]_{2}^{0} - O \right]_{n}^{0} - O$$

wherein. R⁴ is hydrogen or a hydrocarbon group of 1 to 10 carbon atoms; M is a mono- to tri-valent metallic atom; and "n" is an integer of 1 to 3.

[0309] More concretely, they include:

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sodium-bis(4-tert-butylphenyl) phosphate, sodium-bis(4-methylphenyl) phosphate, sodium-bis(4-ethylphenyl) phosphate, sodium-bis(4-i-propylphenyl) phosphate, sodium-bis(4-tert-octylphenyl) phosphate, sodium-bis(4-tert-butylphenyl) phosphate, potassium-bis(4-tert-butylphenyl) phosphate, calcium-bis(4-tert-butylphenyl) phosphate, magnesium-bis(4-tert-butylphenyl) phosphate, lithium-bis(4-tert-butylphenyl) phosphate,

aluminum-bis(4-tert-butylphenyl) phosphate, and a combination thereof. Of these, sodium-bis(4-tert-butylphenyl) phosphate is more preferable.

$$R^5$$
 OH OH

wherein. R⁵ is hydrogen or a hydrocarbon group of 1 to 10 carbon atoms. [0310] More concretely, they include:

- 1,3,2,4-dibenzylidene sorbitol.
- 1,3-benzylidene-2,4-p-methylbenzylidene sorbitol.
- 1,3-benzylidene-2,4-p-ethylbenzylidene sorbitol,
- 1,3-p-methylbenzylidene-2,4-benzylidene sorbitol.
- 1,3-p-ethylbenzylidene-2,4-benzylidene sorbitol.
- 1,3-p-methylbenzylidene-2,4-p-ethylbenzylidene sorbitol,
- 1,3-p-ethylbenzylidene-2,4-p-methylbenzylidene sorbitol,
- 1,3,2,4-di(p-methylbenzylidene) sorbitol,
- 1,3,2,4-di(p-ethylbenzylidene) sorbitol.
- 1,3,2,4-di(p-n-propylbenzylidene) sorbitol,
- 1,3,2,4-di(p-i-propylbenzylidene) sorbitol,
- 1,3,2,4-di(p-n-butylbenzylidene) sorbitol,
- 1,3,2,4-di(p-s-butylbenzylidene) sorbitol,
- 1,3,2,4-di(p-t-butylbenzylidene) sorbitol,
- 1,3,2,4-di(2',4'-dimethylbenzylidene) sorbitol,
- 1,3,2,4-di(p-methoxybenzylidene) sorbitol,
- 1,3,2,4-di(p-ethoxybenzylidene) sorbitol,
- 1,3-benzylidene-2,4-p-chlorobenzylidene sorbitol,
- 1,3-p-chlorobenzylidene-2,4-benzylidene sorbitol,
- 1, 3-p-chlorobenzylidene-2,4-p-methylbenzylidene-sorbitol,
- 1,3-p-chlorobenzylidene-2,4-p-ethylbenzylidene sorbitol,
- 1,3-p-methylbenzylidene-2,4-p-chlorobenzylidene sorbitol,
- 1,3-p-ethylbenzylidene-2,4-p-chlorobenzylidene sorbitol,

1,3,2,4-di(p-chlorobenzylidene) sorbitol, and a combination thereof. Of these, more preferable ones are 1,3,2,4-dibenzylidene sorbitol, 1,3,2,4-di(p-methylbenzylidene) sorbitol, 1,3,2,4-di(p-ethylbenzylidene) sorbitol, 1,3-p-chlorobenzylidene-2,4-p-methylbenzylidene sorbitol, 1,3,2,4-di(p-chlorobenzylidene) sorbitol, and a combination thereof.

[0311] The other nucleating agents useful for the present invention include metallic salts of aromatic and aliphatic carboxylic acids. More concretely, they include aluminum salts of benzoic acid and p-tert-butylbenzoic acid, sodium adipate, sodium thiophenecarboxylate and sodium pyrrolecarboxylate.

[0312] An inorganic compound, e.g., talc, may be used as the nucleating agent.

[0313] The above-described nucleating agent may be blended in copmosition at 0.001 to 10 parts by weight per 100 parts by weight of the propylene-based polymer [C-2], preferably 0.01 to 5 parts by weight, more preferably 0.1 to 3 parts by weight.

[0314] Blend of the nucleating agent can accelerate crystallization of the propylene-based polymer, make the grains finer during the crystallization process, and allow the polymer to be molded faster.

[0315] The propylene-based polymer composition (2) of the present invention can be formed into various shapes, e.g., film, sheet and pipe, by a forming method, e.g., extrusion, injection molding, inflation or calendaring. The formed article is excellent in recovery from the strained conditions.

[0316] The propylene-based polymer composition (2) of the present invention can be widely used for various known purposes of formed polyolefin, e.g., sheet, film (stretched or not stretched), and filament.

[0317] These formed articles are produced by the known thermal forming processes, e.g., extrusion, injection molding, inflation, blow molding, extrusion-blow molding, injection-blow molding, pressing, vacuum molding, calendering and foaming.

[0318] These formed articles are described by several examples.

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[0319] The formed articles related to the present invention are not limited in shape and type, and include sheet, film (not stretched), pipe, hose, wire coating and filament, when they are produced by extrusion. Of these, sheet, film and filament are more preferable.

[0320] The propylene-based polymer composition (2) of the present invention can be extruded by a known extruder under known forming conditions. For example, the molten composition is extruded by a monoaxial screw, kneading, ram or gear extruder through a T-die or the like into a sheet or film (not stretched).

[0321] The above extruded sheet or extruded film (not stretched) can be stretched by a known stretching process, e.g., tenter (longitudinal/transverse or transverse/longitudinal stretching), simultaneous biaxial stretching or monoaxial stretching.

[0322] The sheet or not stretched film is stretched at a stretching ratio normally in a range of around 20 to 70 folds in the case of biaxial stretching, and around 2 to 10 folds in the case of monoaxial stretching. The thickness of the obtainable stretched sheet is preferably around 5 to 200µm.

[0323] An inflation film is one of the film-shaped formed articles for which the propylene-based polymer composition (2) of the present invention can be used. Inflation-formation using the composition (2) makes draw down of moldings difficult

[0324] The formed article of the propylene-based polymer composition (2) of the present invention has various favorable properties, e.g., antistatic property, softness represented by tensile elastic modulus, resistance to heat and scratching, moldability, resistance to aging under heating, transparency, through-view property, gloss, rigidity, moisture-proofness, and gas barrier property. As such, it can be widely used for, e.g., packing film. Its moisture-proofness makes it particularly suitable for press through packs or the like, which is used for packing of tablets and capsules for medicines or the like.

[0325] The filament can be produced by, e.g., extruding the molten propylene-based polymer composition (2) through a spinneret. The filament thus produced may be further stretched, to an extent of molecular orientation at least in one direction, preferably at a stretching ratio normally in a range of around 5 to 10 folds. The filament of the propylene-based polymer composition (2) of the present invention has various favorable properties, e.g., antistatic property, transparency, rigidity, and resistance to heat and impact.

[0326] The propylene-based polymer composition (2) of the present invention can be also injection-molded into various shapes by a known injection molder under known conditions. The injection-molded article of the propylene-based polymer composition (2) of the present invention has various favorable properties, e.g., antistatic property, rigidity, resistance to heat and impact, surface gloss, and resistance to chemicals and wear. As such, it can find wide use, e.g., automobile interiors (e.g., trim), automobile exteriors (e.g., bumper, sidemolding and wheel cover), home electric appliance cases, and containers.

[0327] The propylene-based polymer composition (2) of the present invention can be also blow-molded by a known blow molder under known conditions.

[0328] Taking extrusion-blow molding as an example, the propylene-based polymer composition (2) of the present

invention, molten at 100 to 300°C, is extruded through a die into a tubular parison, which is blown with air while being held in a mold of desired shape, and then put in a mold while being kept at 130 to 300°C, to form the hollow article. It is stretched (blown) preferably at a ratio of around 1.5 to 5 folds in the transverse direction.

[0329] When injection molding is used, the propylene-based polymer composition (2) of the present invention, molten at 100 to 300°C, is injected into a parison mold to form the parison, which is blown with air while being held in a mold of desired shape, and then put in a mold while being kept at 130 to 300°C, to form the hollow article.

[0330] It is stretched (blown) preferably at a ratio of around 1.1 to 1.8 folds in the longitudinal direction and 1.3 to 2.5 folds in the transverse direction.

[0331] The blow-molded article of the propylene-based polymer composition (2) of the present invention is excellent in transparency, rigidity, resistance to heat and scratching, and moldability.

[0332] The pressed article may be represented by the mold-stamped article. For example, when the base and surface materials are simultaneously pressed to form the composite, monolithic article by mold stamping, the propylene-based polymer composition (2) of the present invention can be used for the base.

[0333] The concrete examples of the mold-stamped articles include automobile interiors, e.g., door trim, rear package trim, seat back garnish and instrument panel.

[0334] The propylene-based polymer composition (2) of the present invention is transparent and highly rigid, showing a sufficient rigidity even when it is blended with, e.g., an elastomer. As such, it can find wide use, e.g., automobile interiors (e.g., trim), automobile exteriors (e.g., bumper, sidemolding and wheel cover), home electric appliance cases, and containers.

[0335] The pressed articles of the propylene-based polymer composition (2) of the present invention has various favorable properties, e.g., antistatic property, rigidity, resistance to heat, transparency, resistance to scratching, resistance to aging under heating, surface gloss, and resistance to chemicals and wear.

[0336] The propylene-based polymer composition of the present invention, comprising a propylene-based polymer incorporated with a specific ethylene/ α -olefin copolymer and inorganic filler, has greatly improved balances between rigidity and tensile breaking point elongation, especially these properties at low temperature.

Third propylene-based resin composition (3)

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[0337] The third propylene-based polymer composition (3) of the present invention comprises the propylene-based polymer [C-3], ethylene/ α -olefin copolymer [A], ethylene-based copolymer [B]. In other words, it comprises the propylene-based polymer [C-3] and ethylene/ α -olefin copolymer composition [AB], and hence falls into the category of the propylene-based polymer composition (1).

[0338] The ethylene/ α -olefin copolymer composition [AB] composed of the ethylene/ α -olefin copolymer [A] and ethylene-based copolymer [B] for the second propylene-based polymer composition (2) can be suitably used also for the third propylene-based polymer composition (3).

[0339] For the third propylene-based polymer composition (3) of the present invention, the specific propylene/α-olefin copolymer [C-3] having the following properties is used. The propylene-based copolymer [C-3] may be block or random copolymer so long as it satisfies the following properties, the random copolymer being more preferable.

[0340] The propylene/ α -olefin copolymer [C-3] has a melt flow rate (MFR, determined in accordance with ASTM D1238 at 190°C and a load of 2.16kg) of 0.1 to 400g/10 minutes, preferably 0.1 to 200g/10 minutes.

[0341] The propylene/ α -olefin copolymer having an MFR in the above range can give the propylene/ α -olefin copolymer composition which is excellent in fluidity and can be formed into a large-sized shape. The composition of the copolymer having an MFR more than 400g/10 minutes may have a deteriorated impact resistance (Izod impact strength).

[0342] The propylene/ α -olefin copolymer [C-3] contains at least one type of α -olefin of 2 to 20 carbon atoms other than propylene at 1.5 to 10% by mole, preferably 1.5 to 8%.

[0343] The α -olefins of 2 to 20 carbon atoms other than propylene include ethylene, 1-butene, 1-pentene, 1-hexene, 1-heptene, 1-octene, 1-decene, 1-dodecene, 1-hexadecene and 4-methyl-1-pentene, of which ethylene, 1-butene, 1-hexene and 1-octene are more preferable.

[0344] The propylene/ α -olefin copolymer [C-3] has an intrinsic viscosity [η] of 0.01 to 10dl/g, determined in decalin at 135°C, preferably 0.5 to 6dl/g, more preferably 1.0 to 4dl/g. The copolymer [C-3] having an intrinsic viscosity in the above range can be mixed with another component more easily, and the resultant composition tends to give the formed article of high mechanical strength.

[0345] The propylene/α-olefin copolymer [C-3] preferably has a GPC-determined molecular weight distribution (Mw/Mn) of 6 or less, more preferably-1-5-to-5. The copolymer [C-3] having an Mw/Mn value more than 6 may have deteriorated transparency.

[0346] The propylene/ α -olefin copolymer [C-3] preferably has a glass transition temperature Tg of -10°C or less, measured by a differential scanning calorimeter (DSC), preferably -13°C or less. The copolymer [C-3] having a glass

transition temperature Tg exceeding the above limit may have deteriorated softness.

[0347] The propylene/ α -olefin copolymer [C-3] preferably has a temperature Tm at the maximum peak in the endothermic curve, measured by a differential scanning calorimeter (DSC), i.e., melting point, satisfying the relationship 70 < T_m < 155 - 5.5 (100-P), preferably 90 < T_m < 155 - 5.5 (100-P), wherein P is the propylene content (% by mole) in the copolymer.

[0348] The copolymer [C-3] tends to have excellent transparency and anti-blocking property, when its Tm level satisfy the above relationship.

[0349] It is preferable that the propylene/α-olefin copolymer [C-3] have a microisotacticity of 0.8 or more, more preferably 0.85 or more, with respect to the triad chains of propylene. The copolymer [C-3] having a microisotacticity in the above range tends to have a high crystallization rate and good processability.

[0350] Microisotacticity (hereinafter referred to as triad tacticity or mm fraction) with respect to the triad chains is described.

[0351] Triad tacticity (mm fraction) with respect to the triad chains is defined as intensity (area) ratio of the second side-chain methyl group in the head-tail bonded triad propylene chains, determined by the ¹³C-NMR spectroscopy. It is given by the following relationship (2):

wherein. PPP (mm), PPP (mr) and PPP (rr) are each an area of the second side-chain methyl group in the head-tail bonded triad propylene chains observed by the ¹³C-NMR spectroscopy in each shift region, described below:

First region	Second region	Third region
21.0 to 21.9 ppm	20.3 to 21.0 ppm	19.5 to 20.3 ppm
PPP (mm)	PPP (mr)	

wherein. P is the propylene-derived unit.

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[0352] The PPP (mm), PPP (mr) and PPP (rr) represent the respective head-tail bonded triad propylene chain structures described below:

[0353] In the methyl carbon region (19 to 23ppm), there are the following side-chain methyl group peaks of the propylene unit in the other chains, in addition to those of the head-tail bonded triad propylene chains, and the peak areas of these other types of methyl group are corrected to determine mm fraction by the procedure described below, wherein E is the ethylene-derived unit:

(1) In the second region, there observed a peak derived from the side-chain methyl group, observed in the second-unit (propylene unit) in the PPE triad chain in which propylene is bonded to another propylene in the head-tail bonding manner.

The peak area of the above methyl group can be determined from the peak area of the methine group (reso-

nating at around 30.6ppm) of the second unit (propylene unit) in the PPE triad chain.

(2) In the third region, there observed a peak derived from the side-chain methyl group in the second unit (propylene unit) in the EPE triad chain.

The peak area of the above methyl group can be determined from the peak area of the methine group (resonating at around 32.9ppm) of the second unit (propylene unit) in the EPE triad chain.

(3) In the second and third regions, there observed peaks derived from the methyl groups C to E' in the position-irregular units, shown in the following partial structures (i), (ii) and (iii).

The peaks of the methyl groups C, D and D' are observed in the second region, and those of the methyl groups E and E' in the third region.

[0354] Of the peaks of the methyl groups of the position-irregular units (i), (ii) and (iii), those of the methyl groups A and B, observed at 17.3 and 17.0ppm, respectively, are not found in the first to third regions.

Structure (i)

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Structure (ii)

Structure (iii)

(n ≥ 2)

[0355] The peak area of the methyl group C can be determined from the peak area of the adjacent methine group (resonating at around 31.3ppm).

[0356] The peak area of the methyl group D can be determined by halving totaled areas of the peaks relevant to the $\alpha\beta$ methylene carbon of the structure (ii) (resonating at around 34.3 and 34.5ppm).

[0357] The peak area of the methyl group D' can be determined from the peak area of the methine group (resonating at around 33.3ppm) adjacent to the methyl group E' of the structure (iii).

[0358] The peak area of the methyl group E can be determined from the peak area of the adjacent methine carbon (resonating at around 33.7ppm).

[0359] The peak area of the methyl group E' can be determined from the peak area of the adjacent methine carbon (resonating at around 33.3ppm).

[0360] The peak area of the second side-chain methyl group in the head-tail bonded triad propylene chains can be found by subtracting the above peak areas from the total peak area of the second and third regions.

[0361] Each carbon peak in the spectral pattern can be allocated by referring to the literature (Polymer, 30, 1350 (1989)).

[0362] The propylene-based polymer composition (3) of the present invention comprises the propylene/ α -olefin copolymer [C-3] at 20 to 95% by weight, preferably 30 to 85% by weight, and ethylene/ α -olefin copolymer composition [AB] at 5 to 80%, preferably-15-to-70%. The-ethylene/ α -olefin-copolymer composition [AB] as a modifier-shows-the-sufficient modification effects of improving softness and impact resistance when blended at the upper limit or less, and is less likely to deteriorate the inherent properties of the propylene/ α -olefin copolymer [C-3], e.g., rigidity when incorporated at the lower limit or more.

[0363] When blended in the propylene/ α -olefin copolymer [C-3] at the given content, the ethylene/ α -olefin copolymer composition [AB] as the modifier improves softness and transparency of the polypropylene-based resin while keeping its inherent resistance to heat.

[0364] It is preferable that the propylene/ α -olefin copolymer [C-3] of the present invention has a damping factor (tan δ) peak caused from the glass transition temperature of the propylene/ α -olefin copolymer [C-3] and that caused from the glass transition temperature of the ethylene/ α -olefin copolymer composition [AB], when the temperature-dependence of the modulus of elasticity is measured at intervals of 3°C, and that these peaks are separated from each other.

[0365] The propylene-based polymer composition may be insufficient both in impact resistance and rigidity, when these peaks are not distinctly "separated" but are "fused" with each other.

[0366] The propylene-based polymer composition [C-3] of the present invention may comprise, in addition to the above-described components, one or more additives within limits not harmful to the object of the present invention. The additives useful for the present invention include a nucleating agent, antioxidant, hydrochloric acid absorber, heat stabilizer, light stabilizer, ultraviolet absorber, lubricant, antistatic agent, flame retardant, pigment, dye, dispersant, copper inhibitor, neutralizer, foaming agent, plasticizer, antifoaming agent, crosslinking agent, flow improver (e.g., peroxide), and weld strength improver. Types of these additives and contents in the composition are similar to those described for the second propylene-based polymer composition (2).

[0367] The propylene-based polymer composition (3) of the present invention can be formed into various shapes, e.g., film, sheet and pipe, by a forming method, e.g., extrusion, injection molding, inflation or calendering. The formed article is excellent in recovery from the strained conditions.

[0368] The propylene-based polymer composition (3) of the present invention (sometimes referred to as soft propylene-based polymer composition) can be widely used for various known purposes of formed polyolefin, e.g., sheet, film (stretched or not stretched), and filament, as is the case with the second propylene-based polymer composition (2).

[0369] These formed articles are produced by the known thermal forming processes, e.g., extrusion, injection molding, inflation, blow molding, extrusion-blow molding, injection-blow molding, pressing, vacuum molding, calendering and foaming.

[0370] These formed articles are described by several examples.

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[0371] The formed articles related to the present invention are not limited in shape and type, and include sheet, film (not stretched), pipe, hose, wire coating and filament, when they are produced by extrusion. Of these, sheet, film and filament are more preferable.

[0372] The propylene-based polymer composition (3) of the present invention can be extruded by a known extruder under known forming conditions, as is the case with the second propylene-based polymer composition (2). For example, the molten composition is extruded by a monoaxial screw, kneading, ram or gear extruder through a T-die or the like into a sheet or film (not stretched).

[0373] The above extruded sheet or extruded film (not stretched) can be stretched by a known stretching process, e.g., tenter (longitudinal/transverse or transverse/longitudinal stretching), simultaneous biaxial stretching or monoaxial stretching.

[0374] The sheet or not stretched film is stretched at a stretching ratio normally in a ratio of around 20 to 70 folds in the case of biaxial stretching, and around 2 to 10 folds in the case of monoaxial stretching. The thickness of the obtainable stretched sheet is preferably stretched to a thickness of around 5 to $200\mu m$.

[0375] An inflation film is one of the film-shaped formed articles for which the propylene-based polymer composition (3) of the present invention can be used. Inflation formation using the composition (3) makes draw down of moldings difficult. The composition (3), when inflation-formed, is resistant to draw down.

[0376] The formed article of the propylene-based polymer composition (3) of the present invention, e.g., sheet or film, has various favorable properties, e.g., antistatic property, softness represented by tensile elastic modulus, resistance to heat and scratching, moldability, resistance to aging under heating, transparency, through-view property, gloss, rigidity, moisture-proofness, and gas barrier property. As such, it can be widely used for, e.g., packing film. Its moisture-proofness makes it particularly suitable for press through packs or the like, which is used for packing of, e.g., tablets and capsules for medicines. It is also suitable for medical bags, especially transfusion bags, excellent in gas barrier property, transparency or the like, which are produced by putting the sheets one on another and joining the given portions by a conjugation method, e.g., heat sealing or high-frequency induction heating.

[0377] The filament can be produced by, e.g., extruding the molten propylene-based polymer composition (3) through a spinneret. The filament thus produced may be further stretched, to an extent of molecular orientation at least in one direction, preferably at a stretching ratio normally in a range of around 5 to 10 folds. The filament of the propylene-based polymer composition (3) of the present invention has various favorable properties, e.g., antistatic property, softness, and resistance to heat and impact.

[0378] The propylene-based polymer composition (3) of the present invention can be also injection-molded into various shapes by a known injection molder under known conditions. The injection-molded article of the propylene-based polymer composition (3) of the present invention has various favorable properties, e.g., antistatic property, softness,

resistance to heat and impact, transparency, surface gloss, and resistance to chemicals and wear. As such, it can find wide use, e.g., dress cases, miscellaneous goods, automobile surface materials, and containers such as medical containers.

[0379] The propylene-based polymer composition (3) of the present invention can be also blow-molded by a known blow molder under known conditions.

[0380] Taking extrusion-blow molding as an example, the propylene-based polymer composition (3) of the present invention, in molten state at 100 to 300°C, is extruded through a die into a tubular parison, which is blown with air while being held in a mold of desired shape, and then put in a mold while being kept at 130 to 300°C, to form the hollow article. It is stretched (blown) preferably at a ratio of around 1.5 to 5 folds in the transverse direction.

[0381] When injection-blow molding is used, the propylene-based polymer composition (3) of the present invention, molten at 120 to 300°C, is injected into a parison mold to form the parison, which is blown with air while being held in a mold of desired shape, and then put in a mold while being kept at 130 to 300°C, to form the hollow article.

[0382] Stretched (blown) ratio of the article is preferably in the range of around 1.1 to 1.8 in the longitudinal direction and 1.3 to 2.5 in the transverse direction.

[0383] The blow-molded article made from the propylene-based polymer composition (3) of the present invention is excellent in transparency, softness, resistance to heat and scratching, and moldability.

[0384] The pressed article may be represented by the mold-stamped article. For example, when the substrate and skin materials are simultaneously pressed to form the composite, monolithic article by mold stamping, the propylene-based polymer composition (3) of the present invention can be used for the substrate.

[0385] The concrete examples of the mold-stamped articles include automobile interiors, e.g., door trim, rear package trim, seat back garnish and instrument panel.

[0386] The pressed articles of the propylene-based polymer composition (3) of the present invention has various favorable properties, e.g., antistatic property, softness, resistance to heat, transparency, resistance to scratching, resistance to aging under heating, surface gloss, and resistance to chemicals and wear.

[0387] The third propylene-based polymer composition (3) of the present invention, comprising a specific propylene-based copolymer and ethylene/ α -olefin copolymer composition, can have greatly improved balances with respect to softness, and resistance to heat and impact.

EXAMPLES

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[0388] The present invention is described more concretely by EXAMPLES, which by no means limit the present invention.

[0389] The properties of each resin components were evaluated by the following methods:

1. Properties of the ethylene/α-olefin copolymer [A]

[Density]

[0390] The strand, measured for MFR at 190°C and a load of 2.16kg, was thermally treated at 120°C for 1 hour, and slowly cooled to room temperature in 1 hour, to determine density by the density gradient tube method.

[α -olefin content, and $T\alpha\beta/T\alpha\alpha$, and B value]

[0391] Determined by the ¹³C-NMR spectroscopy.

[Intrinsic viscosity [ŋ]]

[0392] Determined in decalin at 135°C.

50 [Mw/Mn]

[0393] Determined by gel permeation chromatography (GPC) at 140°C with o-dichlorobenzene as the solvent.

[MFR₁₀/MFR₂]

[0394] Melt flow rates MFR₁₀ and MFR₂ were determined in accordance with ASTM D1238 at 190 $^{\circ}$ C and a load of 10kg for the former (MFR₁₀) and 190 $^{\circ}$ C and a load of 2.16kg for the latter (MFR₂), and the ratio was calculated. The higher a MFR₁₀/MFR₂ ratio is, the higher the fluidity of the molten polymer composition becomes, in other words, the

higher the processability of the molten polymer components becomes.

[Glass transition temperature]

[0395] Determined from the endothermic curve, drawn by heating the sample from room temperature at 20°C/minute to 200°C, at which it was held for 5 minutes, then cooling it at 10°C/minute to -150°C, and heating it again at 10°C/minute.

[Crystallinity]

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[0396] Determined by dividing the heat of fusion per unit mass, found from the DSC-determined endothermic peak, by 70cal/g as the heat of fusion of the polyethylene crystal.

2. Properties of the ethylene-based copolymer [C]

[Density]

[0397] The strand, measured for MFR at 190°C and a load of 2.16kg, was thermally treated at 120°C for 1 hour, and slowly cooled to room temperature in 1 hour, to determine density by the density gradient tube method.

[a-olefin content]

[0398] Determined by the ¹³C-NMR spectroscopy.

25 [Melt tension (MT)]

[0399] Determined by measuring the stress in the molten polymer stretched at a constant rate. The pelletized polymer sample was measured by an MT meter (Toyo Seiki Seisakusho) under the conditions of resin temperature: 190°C, extrusion speed: 15mm/minute, winding speed: 10 to 20m/minute, nozzle diameter: 2.09mm and nozzle length: 8mm.

[MFR]

[0400] MFR₂ was determined in accordance with ASTM D1238 at a given temperature and load of 2.16kg.

35 [Softening temperature (T_m)]

[0401] Tm was regarded as the maximum peak temperature in the endothermic curve determined by a DSC. Measurement is conducted by heating the sample put in an aluminum pan at 10°C/minute to 200°C, at which it was held for 5 minutes, then cooling it at 20°C/minute to room temperature, and heating it again at 10°C/minute, thereby determining the endothermic curve.

3. Properties of the ethylene/α-olefin copolymer composition and propylene-based copolymer composition

[Bending modulus of elasticity (FM)]

[0402] Determined in accordance with ASTM D790.

Test piece: 12.7mm wide, 3.2mm thick, and 127mm long

Span: 64mm

Bending speed: 20mm/minute

[Tensile modulus of clasticity]

[0403] Determined in accordance with JIS K6301.

Span: 30mm Temperature: 23°C

Stretching speed: 30mm/minute

[Izod impact strength (IZ)]

[0404] Determined in accordance with ASTM D256 with the following test piece:

Temperature: 23°C and -30°C

Test piece: 12.7mm wide, 6.4mm thick, and 64mm long

Notch: Provided by machining

[Rockwell hardness]

[0405] Determined in accordance with ASTM D785 with the following test piece:

Test piece: 110mm square (110mm longitudinal imes 110mm transversal) and 3 mm long

R scale

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Temperature: 23°C

[JIS A hardness]

[0406] Determined in accordance with JIS K7215 with the following test piece prepared by pressing:

Test piece. 35mm square (35mm longitudinal imes 35mm transversal) and 3 mm long

Temperature: 23°C

[Viscoclasticity]

[0407] Temperature-dependence of dynamic viscoelasticity was measured by a Rheotometrics's RDS II at a frequency of 62.5rad/second in a temperature range from -80 to 50°C, to judge whether the damping factor ($\tan\delta$) peak derived from the glass transition temperature of the propylene-based copolymer [C] and that derived from the glass transition temperature of the ethylene/ α -olefin copolymer composition are separated or fused.

PRODUCTION EXAMPLE 1

[0408] 845ml of hexane kept at 23°C was put in a 2-liter capacity stainless autoclave equipped with an agitator was sufficiently purged with nitrogen. 155ml of 1-butene was added with agitation, while it was cooled with ice water, to the autoclave. Then, the autoclave was heated to 60°C inside, and pressurized with ethylene to 8kg/cm² as the total pressure. When the inside pressure reached 8kg/cm², 1.0ml of a 1.0mM/ml decane solution of triisobutyl aluminum (TIBA) was carried by nitrogen into the autoclave under pressure, and then 0.3ml of a toluene solution prepared beforehand to contain 0.3mM (as Al) of methyl aluminoxane and 0.001mM of rac-dimethylsilylene-bis[1-(2-methyl-4-phenyl-indenyl)] zirconium dichloride was carried by nitrogen into the autoclave under pressure, to start the polymerization.

[0409] Temperature in the autoclave was controlled at 60°C and ethylene was directly charged to keep pressure at 8kg/cm² in the autoclave for 30 minutes. Methanol (5ml) was pumped into the autoclave 30 minutes after the polymerization was started, to terminate the polymerization, and pressure in the autoclave was released to the atmospheric pressure. Then, 2-liter of acetone was added to the effluent.

[0410] The solvent-containing, rubber ball-shaped polymer thus prepared was dried under the conditions of 130°C and 600torr for 13 hours. This produced 47g of ethylene/1-butene copolymer containing 39mmol of 1-butene.

[0411] Table 1 gives the basic properties of the ethylene/1-butene copolymer (A1) thus prepared. The ethylene/ α -olefin copolymers (A2) to (A4), ethylene-based copolymers (B1) to (B5), and propylene-based polymer (C1) were prepared in the same manner as in the above, except that the monomer types and charge quantities were changed to have the copolymer compositions shown in Table 1. Table 1 also gives the basic properties of the ethylene/ α -olefin copolymers (A2) to (A4), ethylene-based copolymers (B1) to (B5), and propylene-based polymer (C1).

5÷		2	1-hexene 1-octene 1-butene 1-octene 1-butene 1-hexene 1-hexene propylene
10		85	1-hexene
15		. B4	1-hexene
20		B3	1-butene
25		82	1-octene
20	Table 1	B1	1-butene
30	Tat	A4	1-butene
35		A3	1-octene
10		A2	1-hexene
15		A1	1-butene
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Copotymers	A1	A2	A3	A4	B1	B2	B3	B4	BS	2
α-olefin types	1-butene	1-hexene	1-octene	1-butene	1-butene	1-octene	1-butene	1-hexene	1-hexene	propylene
Ethylene contents (mol)	61	99	99	30	84	84	06	95.2	98.6	not included
Density (g/cm)	98'0'	<u>9</u> 96'0 .	0.855	0.861	0.872	0.873	0.89	0.908	0.938	0.91
[n]	, 2.3	2.2	2.1	2.2	4.1	1.4	1.4	1.4	12	
Mw/Mn	2.4	2.3	2.3	2.5						
MFR (g/10min)					4.1	4.3	4.2	1.7	4.5	55
MFR ₁₀ /MFR ₂	10.2	6.6	9.6	10.5						
MT(g)					0.5	0.4	9.0	1.2	0.4	
Tg(°C)	-64	-68	69-	-59	-59	-62	-53			
Crystallinity (%)	0 ,	0	0	0						
тт(°С)					99	64	84	94	118	167
400×d-250					66	66	106	113	125	131
Ταβ/Ταα	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01
B value	1.11	1.07	1.08	-	1.03	1.04	-	-	1.03	1.04

EXAMPLE 1

[0412] A mixture of 20% by weight of the ethylene/1-butene copolymer (A1) and 80% by weight of the ethylene-based copolymer (B1) was kneaded by a Lab-plastomill (Toyo Seiki, Ltd.) at 60rpm and 200°C for 5 minutes and pelletized, to have the pre-blend pellets of the ethylene/ α -olefin copolymer composition. The pellets were subjected to the pellet blocking test, to evaluate the blocking property. The result is given in Table 1.

[Pellet blocking test]

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- [0413] The pellets, put in a polyethylene bag, were allowed to stand under a load of 100g/cm² for 72 hours in an oven kept at 35°C, and withdrawn to observe the blocking conditions. They were rated by the following marks:
 - (ii) : Blocking is rarely observed.
 - O: Blocking is observed, but the agglomerated pellets can be easily loosened by the finger.
 - Δ: Blocking is observed, but the agglomerated pellets can be loosened, when pressed by the finger.
 - \times : The pellets are fused with each other into a veil-like condition.

EXAMPLES 2 to 7

20 [0414] The pellets of the ethylene/α-olefin copolymer compositions were prepared in the same manner as in EXAM-PLE 1, except for the compositions shown in Table 2, and subjected to the pellet blocking tests also in the same manner as in EXAMPLE 1. The pellets of each composition were resistant to blocking and handled easily.

[0415] The pellets prepared in EXAMPLES 3 to 7 were also measured for the component soluble in decane at room temperature (% by weight), ethylene content and softening point (T_m) .

[0416] The results are given in Table 2.

COMPARATIVE EXAMPLES 1 to 5

[0417] The pellets were prepared in the same manner as in EXAMPLE 1, except for the compositions shown in Table 2, and measured for MFR, blocking property, the component soluble in decane at room temperature (% by weight), ethylene content, the component soluble in decane at room temperature (% by weight) and softening temperature (T_m). The pellets prepared in each of the comparative examples showed blocking, and were difficult to handle.

[0418] The results are given in Table 2.

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Table 2

Comp Ex. 55				901						0.3	00_	36		×
Comp Ex. 4			5							0.5	100	99	ı	×
Comp Ex. 3		9								0.5	100	99	1	×
Comp Ex. 2	<u>6</u>									0.4	100	61	1	×
Comp Ex. 1				20	80					2,4	. 58	44	56	0
Example Example Example Example Example Example 7			20		8					2.7	22	69	56	0
Example 6		20			80					2.7	22	69	56	0
Example 5	20								8	2.8	20	61	1	0
Example 4	20		-			·		8		1.3	50	61	84	0
Example 3	20						80			2.6	20	62	89	0
Example 2	20					80				2.7				0
Example 1	20				80					2.6				0
	A1	A2	A3	A4	B1	B2	83	84	98	MFR(g∕10min)	Content (% by weight)	Ethylene content (% by mole)	Tπ(°C)	Blocking property
					compositions					MFR(g.	Component soluble in	decane at room Ethylene temperature (% by mole)	Tm	Blocking

EXAMPLE 8

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[0419] A mixture of 20% by weight of the ethylene/1-butene copolymer (A1) and 80% by weight of the ethylene-based copolymer (B1) was kneaded by a Lab-plastomill (Toyo Seiki, Ltd.) at 60rpm and 200°C for 5 minutes and pelletized, to have the pre-blend pellets. Then, 30% by weight of the pre-blend pellets of the ethylene/α-olefin copolymer/ethylene-based copolymer composition was incorporated with 70% by weight of the propylene-based polymer (C1), and 0.1% by weight of calcium stearate, 0.1% by weight of Inganox 1010 and 0.1% by weight of Inganox 168 as the stabilizers. The mixture was kneaded by a Lab-plastomill (Toyo Seiki, Ltd.) at 150rpm and 200°C for 5 minutes and pelletized.

[0420] The pellets were pressed at 230°C, and measured for breaking point strength, bending modulus of elasticity, impact strength and viscoelasticity by the following methods.

[Breaking point strength (TS)]

[0421] Determined in accordance with ASTM D638 at room temperature.

[Bending modulus of elasticity]

[0422] Determined in accordance with ASTM D790 with the 2mm thick test piece injection-molded under given conditions under the conditions of span: 32mm and bending speed: 5m/minute.

[Impact strength (IZ)]

[0423] Determined in accordance with ASTM D256 with the 3mm thick test piece (rear notch) at -30°C.

[Viscoelasticity]

[0424] Temperature-dependence of dynamic viscoelasticity was measured by a Rheotometrics's RDS II at a frequency of 62.5rad/second in a temperature range from -80 to 50°C, to judge whether the damping factor ($\tan \delta$) peak derived from the glass transition temperature of the propylene-based copolymer [C] and that derived from the glass transition temperature of the ethylene/ α -olefin copolymer composition are separated or fused.

EXAMPLES 9 to 12

35 [0425] The formed articles were prepared in the same manner as in EXAMPLE 8, except that the ethylene/α-olefin copolymer and ethylene-based polymer shown in Table 3 were used for each of the examples, and were evaluated also in the same manner as in EXAMPLE 8.

[0426] The results are given in Table 3.

40 COMPARATIVE EXAMPLES 6 to 11

[0427] The formed articles were prepared in the same manner as in EXAMPLE 8, except that the ethylene/ α -olefin copolymer and ethylene-based polymer shown in Table 3 were used for each of the examples, and were evaluated also in the same manner as in EXAMPLE 8.

⁴⁵ [0428] The results are given in Table 3.

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Table

		Example	Example	Example 1	Example Example Example 1 Example 1	Exa	Сощ	Comp	Comp	Сошр	Сошр	Сошр
		8	S)	Э.			Ex. 6	Ex. 7	 ∦	Ex.	Ex. 10	Ex. 11
Ethylene/a-	A1	9			9	9	9	9		30		
olefin	A2		9								စ္တ	
conolymer	A3			9								30
in the day	. A4								9			
	. 81	24	24	24					24			
Ett. damente					24							
culylene - pased	B3					24						
-	B4						24					
	. 85							24	-			
Propyle copoly	Propylene-based copolymer C1	07	07	70	7.0	70	70	70	70	70	70	5
Component	Content (% by weight)					6.1		9		30		
decane at room	Ethy											
temperature	content (% by mole)					62		8		9		
Component	Content					30.8		8		30.6		
soluble in	(% by weight)							25		2.2		
decane at 63°C	Tm(°C)					89		ı		* 1		
	TS(MPa)	21	23	23	25	27	28	29.5	20	18	18	18
Characteristics	FM(1300	1350	1380	1350	1400	1450	1530	1200	800	780	750
	(m/f)ZI	8	33	35	35	35	.20	15	20	22	24	. 25
	lan O peaks Separated Separated Separated	Separated	Separated	Separated	Separated	Separated Separated Separated Fused	Separated	Separated	Separated		Fused	Filsed

[0429] As shown in Table 3, it is apparent that the propylene-based polymer compositions prepared in EXAMPLES -8-to-12-were-well-balanced with respect to breaking point strength; bending modulus of elasticity and impact strength. [0430] The formed articles of the compositions prepared in COMPARATIVE EXAMPLES 6 and 7 (which used the ethylene-based copolymer [8] of high density) and that of the one prepared in COMPARATIVE EXAMPLE 8 (which used the ethylene/ α -olefin-based copolymer [A] containing a high content of the α -olefin) were not well-balanced with

respect to impact strength and bending modulus of elasticity, i.e., low impact strength for a high bending modulus of elasticity.

[0431] The formed articles of the compositions prepared in COMPARATIVE EXAMPLES 9 to 11, containing no ethylene-based copolymer [B], were low both in impact strength and bending modulus of elasticity.

EXAMPLE A1

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[0432] A mixture of 20% by weight of the ethylene/1-butene copolymer (A1), prepared in PRODUCTION EXAMPLE 1, and 80% by weight of the ethylene-based copolymer (B1) was kneaded by a Lab-plastomill (Toyo Seiki, Ltd.) at 60rpm and 200°C for 5 minutes and pelletized, to obtain the pre-blend pellets of the ethylene/ α -olefin copolymer/ ethylene-based copolymer composition (AB-1). The pre-blend pellets were evaluated for tensile modulus of elasticity and JIS A hardness. The result is given in Table 4.

EXAMPLES A2 to A-5, and COMPARATIVE EXAMPLES A1 and A2

[0433] The pre-blend pellets (AB-2) to (AB-7) were prepared in the same manner as in EXAMPLE 13, except for the compositions shown in Table 4, and were evaluated for tensile modulus of elasticity and JIS A hardness. The result is given in Table 4.

Table 4

				Table 4				
		Ex. A1	Ex. A2	Ex. A3	Ex. A4	Ex. A5	Comp Ex. A1	Comp Ex. A2
Pellet composit	tions	AB-1	AB-2	AB-3	AB-4	AB-5	AB-6	AB-7
Compositions	A1	20	20	20			20	
	A2				20			
	А3					20		***
	A4							20
	B1	80						
	B2		80		80	80		80
	B3			80		-		
	B4						80	
Density ratio (da	2/d1)	1.01	1.02	1.03	1.02	1.02	1.06	1.01
Tensile modulus of (MPa)	elasticity	3.6	3.8	15.1	3.2	3.3	20.3	1.5
JIS A hardne	ss	54	56	73	54	54	77	45

EXAMPLE A6

[0434] A molton mixture of 24% by weight of the pellets of the ethylene/1-butene copolymer composition (AB-1), prepared in EXAMPLE A1, 66% by weight of the homopolypropylene (C1), 10% by weight of talc, and 0.1% by weight of calcium stearate, 0.1% by weight.of Inganox 1010 and 0.1% by weight of Inganox 168 as the stabilizers was kneaded by a biaxial extruder at 200°C, and pelletized. The pellets were injection-molded by a 55 tons injection molder (Toshiba Kikai) at 200°C as cylinder temperature and 40°C as mold temperature, and were evaluated for their properties.

[0435] The polypropylene (C1) and talc used in this example are described in Table 5.

Table 5

C₁ C2 Polypropylene Block Polypropylene types types polypropylene MFR(g/10min) 55 65 Component soluble in decane at room temperature 0.7 10.5 (% by weight) Intrinsic viscosity [h] (dl/g) of the component soluble in decane 0.7 8.2 at room temperature I5 of the component soluble 0.982 0.976 in decane at room temperature MFR (g/10 minutes) of the component soluble in decane 135 at room temperature

	(D) Talc
Average particle size (mm)	2.5

[0436] The results are given in Table 6.

EXAMPLE A7

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[0437] The pellets were formed in the same manner as in EXAMPLE A6, except that the ethylene/1-butene copolymer composition (AB-1) was replaced by the ethylene/ α -olefin copolymer composition (AB-2) of different α -olefin content, and were evaluated for their properties.

[0438] The results are given in Table 6.

EXAMPLE A8

40 [0439] The pellets were formed in the same manner as in EXAMPLE A6, except that the ethylene/1-butene copolymer composition (AB-1) was replaced by the ethylene/α-olefin copolymer composition (AB-3) of different α-olefin type and content, and were evaluated for their properties.

[0440] The results are given in Table 6.

45 EXAMPLE A9

[0441] The pellets were formed in the same manner as in EXAMPLE A6, except that the ethylene/1-butene copolymer composition (AB-1) was replaced by the ethylene/ α -olefin copolymer composition (AB-4) of different α -olefin type and content, and were evaluated for their properties.

[0442] The results are given in Table 6.

EXAMPLE A10

[0443] The pellets were formed in the same manner as in EXAMPLE A6, except that the ethylene/1-butene copolymer-composition (AB+1) was replaced by the ethylene/ α -olefin copolymer composition (AB+5) of different α -olefin type and content, and were evaluated for their properties.

[0444] The results are given in Table 6.

EXAMPLE A11

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[0445] A molten mixture of 25% by weight of the pellets of the ethylene/1-butene copolymer composition (AB-1), prepared in EXAMPLE A1, 29% by weight of the homopolypropylene (C1), 36% by weight of the block polypropylene (C2), 10% by weight of talc, and 0.1% by weight of calcium stearate, 0.1% by weight of Inganox 1010 and 0.1% by weight of Inganox 168 as the stabilizers was kneaded by a biaxial extruder at 200°C, and pelletized. The pellets were injection-molded by a 55t injection molder (Toshiba Kikai) at 200°C as cylinder temperature and 40°C as mold temperature, and were evaluated for their properties.

[0446] The results are given in Table 6.

COMPARATIVE EXAMPLE A1

[0447] The pellets were formed in the same manner as in EXAMPLE A6, except that the ethylene/1-butene copolymer composition (AB-1) was replaced by the ethylene/ α -olefin copolymer composition (AB-6) of different α -olefin type and content, and were evaluated for their properties.

[0448] The results are given in Table 6.

COMPARATIVE EXAMPLE A2

20 [0449] The pellets were formed in the same manner as in EXAMPLE A6, except that the ethylene/1-butene copolymer composition (AB-1) was replaced by the ethylene/α-olefin copolymer composition (AB-7) of different α-olefin type and content, and were evaluated for their properties.

[0450] The results are given in Table 6.

COMPARATIVE EXAMPLE A3

[0451] The pellets were formed in the same manner as in EXAMPLE A6, except that the ethylene/1-butene copolymer composition (AB-1) was replaced by the ethylene/ α -olefin copolymer (A1) itself, and were evaluated for their properties. [0452] The results are given in Table 6.

COMPARATIVE EXAMPLE A4

[0453] The pellets were formed in the same manner as in EXAMPLE A11, except that the ethylene/1-butene copolymer composition (AB-1) was replaced by the ethylene/ α -olefin copolymer composition (B-7) of different α -olefin type and content, and were evaluated for their properties.

[0454] The results are given in Table 6.

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 Separated Separ

IZ(J/m)(-30°C) Rockwell hardness

tan ô peaks

		-															
5			es	A4							25		29	36	01.	1550	580
<u>.</u>			ய	A3								24	99		2	1800	200
10			Comparative	A2					,		24		99		10	1850	270
7-1 15				A1						24			.09		10	2020	190
20				A11	25								29	36	10	1650	640
				A10					24				99		10	2000	220
25	ole 6		ples	A9				24					98		10	2000	230
30	Table		Examples	Α8			24						99		10	2010	200
35				A7		24							99		10	1990	250
40				A6	24								99		10	2000	220
				,	AB-1	AB-2	AB-3	AB-4	AB-5	AB-6	AB-7	efin 1)	0.1	C2			
45 50	40. 4.			,			Ethylene/α-olefin	copolymer	compositions			Ethylene/α-olefin copolymer (A-1)	Drocylene	allalydol	Talc (D)	FM(MPa)	IZ(J/m)(53°C)
										Compo-	SICIONS						

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PRODUCTION EXAMPLE 2

[0455] The propylene/ethylene random copolymer (C3) was prepared in the presence of a known solid titanium catalyst. It had an intrinsic viscosity $[\eta]$ of 2.0dl/g, α -olefin containing the ethylene-derived unit at 3.4% by mole and butene-1-derived unit at 2.9% by mole, melting point of 130°C, microisotacticity of 0.97, molecular weight distribution (Mw/Mn) of 4.1 and glass transition temperature Tg of -19°C.

[0456] Table 7 describes the properties of the propylene/ethylene random copolymer (C3) used for the present invention.

	Table 7	
	СЗ	
Polypropylene type	Polypropylene ra	ndom copolymer
MFR (g/10 min)	6.4	•
α-olefin types	Ethylene	Butene-1
α-olefin contents (mol %)	3.4	2.9
Tm (°C)	130	
Mw/Mn	4.1	
Tg (°C)	-19	
[η]	2.0	
Microisotacticity	0.97	

EXAMPLE B1

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[0457] A molten mixture of 40% by weight of the pellets of the ethylene/1-butene copolymer composition (AB-1), prepared in EXAMPLE A1. 60% by weight of the propylene/ethylene random copolymer (C3), and 0.1% by weight of calcium stearate, 0.1% by weight of Inganox 1010 and 0.1% by weight of Inganox 168 as the stabilizers was kneaded by a biaxial extruder at 200°C, and pelletized. The pellets were injection-molded by a 55 tons injection molder (Toshiba Kikai) at 200°C as cylinder temperature and 40°C as mold temperature, and were evaluated for their properties. The test pieces prepared by pressing at 200°C were used for evaluation of haze, TMA and viscoelasticity.

[0458] The results are given in Table 8.

EXAMPLE B2

[0459] The pellets were formed in the same manner as in EXAMPLE B1, except that the ethylene/1-butene copolymer composition (AB-1) was replaced by the ethylene/ α -olefin copolymer composition (AB-2) of different α -olefin content, and were evaluated for their properties.

[0460] The results are given in Table 8.

EXAMPLE B3

[0461] The pellets were formed in the same manner as in EXAMPLE B1 except that the ethylene/1-butene copolymer composition (AB-1) was replaced by the ethylene/ α -olefin copolymer composition (AB-3) of different α -olefin type and content, and were evaluated for their properties.

[0462] The results are given in Table 8.

EXAMPLE B4

[0463] The pellets were formed in the same manner as in EXAMPLE B1, except that the ethylene/1-butene copolymer composition (AB-1) was replaced by the ethylene/ α -olefin copolymer composition (AB-4) of different α -olefin type and content, and were evaluated for their properties.

[0464] The results are given in Table 8.

EXAMPLE B5

[0465] The pellets were formed in the same manner as in EXAMPLE B1, except that the ethylene/1-butene copolymer composition (AB-1) was replaced by the ethylene/α-olefin copolymer composition (AB-5) of different α-olefin type and content, and were evaluated for their properties.

[0466] The results are given in Table 8.

COMPARATIVE EXAMPLE B1

[0467] The pellets were formed in the same manner as in EXAMPLE B1, except that the ethylene/1-butene copolymer composition (AB-1) was replaced by the ethylene/α-olefin copolymer composition (AB-6) of different α-olefin type and content, and were evaluated for their properties.
 [0468] The results are given in Table 8.

COMPARATIVE EXAMPLE B2

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[0469] The pellets were formed in the same manner as in EXAMPLE B1, except that the ethylene/1-butene copolymer composition (AB-1) was replaced by the ethylene/ α -olefin copolymer composition (AB-7) of different α -olefin type and content, and were evaluated for their properties.

[0470] The results are given in Table 8.

COMPARATIVE EXAMPLE B3

[0471] The pellets were formed in the same manner as in EXAMPLE B1, except that the ethylene/1-butene copolymer composition (AB-1) was replaced by the ethylene/α-olefin copolymer (A1) itself, and were evaluated for their properties.

[0472] The results are given in Table 8.

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Table 8

		- 			Examples			Comparative Examples	ive Exa	mples
			81	82	83	84	8.5	18	B2	83
		AB-1	40							
		AB-2		40			.			
	Ethylene/α-olefin	AB-3			40					
-	copolymer	AB-4				40				
	compositions	AB-5					40.			
		AB-6						40		
		AB-7							40	
	Ethylene/a-olefin copolymer	lymer								40
	Propylene-based copolymer	Vmer								I
	(03)		09	09	60	09	09	09	09	09
	FM(MPa)		370	380	400	360	350	200	310	290
	Наге		25	22	18	23	24	39	20	17
	IZ(J/m)(-10°C)		a B	8 N	B N	82	82	340	82	8 R
	TMA(°C)		120	120	120	120	120	120	105	5
	tan ô poaks		Separated	Separated	Separated Separated Separated Separated Separated Fused Fused	Separated	Separated	Separated	Fused	Fused

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Claims

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- 1. An ethylene-based copolymer [I] characterized by:
 - (a-1) having temperature (T_m) at the maximum peak in the endothermic curve, measured by a differential scanning calorimeter (DSC), of 40 to 90°C,
 - (a-2) containing the component soluble in decane at normal temperature at 1 to 70% by weight, and
 - (a-3) containing the component soluble in decane at normal temperature which comprises (i) a recurring unit derived from ethylene and (ii) a recurring unit derived from an α -olefin of 4 or more carbon atoms, and in which a content of (i) the recurring unit derived from ethylene is 50 to 75% by mole.
- 2. An ethylene-based copolymer composition [II] characterized by containing the ethylene-based copolymer which satisfies the following conditions (b-1) to (b-3):
- (b-1) having temperature (T_m) at the maximum peak in the endothermic curve of the component soluble in decane at 64°C, measured by a differential scanning calorimeter (DSC), of 40 to 90°C.
 - (b-2) containing the component soluble in decane at 64°C which contains the component soluble in decane at normal temperature at 1 to 70% by weight, and
 - (b-3) containing the component soluble in decane at normal temperature which comprises (i) a recurring unit derived from ethylene and (ii) a recurring unit derived from an α -olefin of 4 or more carbon atoms, and in which a content of (i) the recurring unit derived from ethylene is 50 to 75% by mole.
 - 3. A propylene-based copolymer composition [III] characterized by containing the propylene-based copolymer and satisfying the following conditions (c-1) to (c-3):
 - (c-1) containing the component soluble in decane at 64°C at 1 to 99% by weight which has a temperature (T_m) at the maximum peak in the endothermic curve, measured by a differential scanning calorimeter (DSC), of 40 to 90°C,
 - (c-2) containing the component soluble in decane at 64°C which contains the component soluble in decane at normal temperature at 1 to 70% by weight, and
 - (c-3) containing the component soluble in decane at normal temperature which comprises (i) a recurring unit derived from ethylene and (ii) a recurring unit derived from an α -olefin of 4 or more carbon atoms, and in which a content of (i) the recurring unit derived from ethylene is 50 to 75% by mole.
- 35 4. An ethylene/ α -olefin copolymer composition comprising:
 - 1 to 70% by weight of [A] a copolymer of ethylene and α -olefin of 4 to 20 carbon atoms which comprises (Ai) 50 to 70% by mole of (a) a recurring unit derived from ethylene and 30 to 50% by mole of (b) a recurring unit derived from an α -olefin of 4 to 20 carbon atoms, and
 - 30 to 99% by weight of [B] an at least one of ethylene-based copolymer comprising (a) ethylene, (b) at least one selected from α -olefin of 4 to 20 carbon atoms and cyclic olefin-based compound, and

having (B-i) a density of 0.870 to 0.895g/cm³.

- The ethylene/α-olefin copolymer composition according to Claim 4 which has a melt flow rate (MFR at 190°C and a load of 2.16kg) of 0.01 to 50g/10 minutes.
 - The ethylene/α-olefin copolymer composition according to Claim 4 or 5, wherein ratio of density (d_b) of said ethylene-based copolymer [B] to density (d_a) of said ethylene/α-olefin copolymer [A] (d_b/d_a) is 1.05 or less.
 - The ethylene/α-olefin copolymer composition according to one of Claims 4 to 6 which has (A-ii) a glass transition temperature of -60°C or lower, determined by a differential scanning calorimeter (DSC) and crystallinity of 1% or less
- 55 --8.—The-ethylene/α-olefin copolymer composition according to one of Claims 4 to 7 which has (A-iii) a B value of 0.9 to 1.5, given by the following general formula (1):

$$B value = [P_{0E}]/(2[P_E][P_0])$$
 (1)

- wherein, $[P_E]$ is molar fraction of the ethylene-derived recurring unit in the copolymer, $[P_0]$ is molar fraction of the α -olefin-derived recurring unit in the copolymer, and $[P_{0E}]$ is a number ratio of the ethylene/ α -olefin copolymer chains to the total dyad chains in the copolymer, determined by the ¹³C-NMR spectroscopy.
- 9. The ethylene/ α -olefin copolymer composition according to one of Claims 4 to 8 which has (A-iv) an intrinsic viscosity [η] of 0.1 to 10.0dl/g, determined in decalin at 135°C.
- 10. The ethylene/α-olefin copolymer composition according to one of Claims 4 to 9, wherein said ethylene-based copolymer [B] has a melt flow rate (190°C and a load of 2.16kg) of 0.1 to 50g/10 minutes.
- 11. The ethylene/α-olefin copolymer composition according to one of Claims 4 to 10, wherein said ethylene-based copolymer [B] has (B-iii) a temperature (T_m) at the maximum peak in the endothermic curve, measured by a differential scanning calorimeter (DSC), correlated with density (d) by the following relationship:

$Tm < 400 \times d-250$

- 12. A resin modifier composed of the ethylene/ α -olefin copolymer composition according to one of Claims 4 to 11.
- 13. A method for modifying resin **characterized by** blending the pellets of the ethylene/α-olefin copolymer composition according to one of Claims 4 to 11 and resin to be modified in the molten state.
- 14. A propylene-based polymer composition comprised of the propylene-based polymer [C-1], ethylene/α-olefin co-polymer [A] and ethylene-based copolymer [B], and characterized by:
 - (i) containing said propylene-based polymer [C-1] at 99 to 1% by weight, and ethylene/α-olefin copolymer [A] and ethylene-based copolymer [B] at 1 to 99% by weight (total content of [A] and [B]), and
 - (ii) having a content ratio of the ethylene/ α -olefin copolymer [A] to the ethylene-based copolymer [B], i.e., [A] /[BJ content ratio, is 1/99 to 70/30.
- 15. The propylene-based polymer composition according to Claim 14, wherein said propylene-based polymer [C-1] is contained at 98 to 60% by weight, and said ethylene/α-olefin copolymer [A] and ethylene-based copolymer [B] at 2 to 40% by weight (total content of [A] and [B]).
- 16. The propylene-based polymer composition according to Claim 14 or 15, wherein said propylene-based polymer [C-1] has a melt flow rate (230°C and a load of 2.16kg) of 0.01g/10 minutes or more.
- 17. The propylene-based polymer composition according to one of Claims 14 to 16 which has a damping factor peak caused from the glass transition temperature of said propylene-based polymer [C-1] and that caused from the glass transition temperature of said ethylene/α-olefin copolymer composition [AB] (composed of said ethylene/α-olefin copolymer [A] and ethylene-based copolymer [B]), when the temperature-dependence of the modulus of elasticity is measured, wherein these peaks are separated from each other.
- 18. A propylene-based copolymer composition comprised of 20 to 93% by weight of [C-2], 6 to 79% by weight of [AB] and 1 to 25% by weight of [D], wherein:

[C-2]: a propylene-based polymer:

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- (1) having a melt flow rate (MFR, at 230°C and a load of 2.16kg) of 0.1 to 400g/10 minutes, determined in accordance with ASTM D-1238,
- (2) containing 0.01 to 30% by weight of the component soluble in decane at normal temperature, and an ___intrinsic-viscosity-[η]-of-0.2-to-10dl/g, determined in decalin at-135°C, and _____
 - (3) having a pentad isotacticity (I_5) of 0.95 or more, determined by the ¹³C-NMR spectroscopy for the component insoluble in decane at normal temperature;

[AB]: a composition composed of:

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[A-1]: a copolymer of ethylene and α -olefin of 4 to 20 carbon atoms,

- (i) comprising 50 to 70% by mole of (a) a recurring unit derived from the ethylene and 30 to 50% by mole of (b) a recurring unit derived from the α -olefin of 4 to 20 carbon atoms, and
- (ii) having an intrinsic viscosity [η] of 0.1 to 10.0dl/g, determined in decalin at 135°C, and

[B-1]: a copolymer of (a) ethylene and (b) at least one compound selected from the group consisting of α -olefins and cyclo-olefins of 3 to 20 carbon atoms, having

- (i) a density of 0.870 to 0.895g/cm³ and
- (ii) a melt flow rate (190°C and a load of 2.16kg) of 0.1 to 50g/10 minutes,

wherein the copolymers [A-1] and [B-1] are incorporated at 1 to 50% and 50 to 99% by weight, respectively, based on the whole composition [AB], and [D]: an inorganic filler.

- 19. The propylene-based copolymer composition according to Claim 18, wherein ratio of density (d₂) of said ethylene-based copolymer [B-1] to density (d₁) of said ethylene/α-olefin copolymer [A-1] (d₂/d₁) is 1.05 or less.
- 20. The propylene-based copolymer composition according to Claim 18 or 19 which has a damping factor peak caused from the glass transition temperature of said propylene-based polymer [C-2] and that caused from the glass transition temperature of said ethylene/α-olefin copolymer composition [AB], when the temperature-dependence of the modulus of elasticity is measured, wherein these peaks are separated from each other.
- 21. A formed article of the propylene-based copolymer composition according to one of Claims 18 to 20.
- 22. The formed article according to Claim 21 which is a sheet or injection-molded article.
- 23. The formed article according to Claim 21 which is an automobile interior or exterior member, or electric appliance case.
- 24. A propylene-based copolymer composition composed of 20 to 95% by weight of [C-3] and 5 to 80% by weight of [AB-1], wherein:

[C-3]: a propylene/α-olefin copolymer:

- (1) having an intrinsic viscosity [η] of 0.01 to 10dl/g, determined in decalin at 135°C, and
- (2) comprising propylene and 1.5 to 10% by mole of an α -olefin, other than propylene, of 2 to 20 carbon atoms. and

[AB-1]: an ethylene/ α -olefin copolymer composition, composed of:

- [A-1]: an ethylene/ α -olefin copolymer of ethylene and α -olefin of 4 to 20 carbon atoms,
 - (i) comprising 50 to 70% by mole of (a) a recurring unit derived from the ethylene and 30 to 50% by mole of (b) a recurring unit derived from the α -olefin of 4 to 20 carbon atoms, and
 - (ii) having an intrinsic viscosity $[\eta]$ of 0.1 to 10.0dl/g, determined in decalin at 135°C, and

[B-1]: an ethylene-based copolymer of (a) ethylene and (b) at least one compound selected from the group consisting of α -olefins and cyclo-olefins of 3 to 20 carbon atoms, having

- (i) a density of 0.870 to 0.895g/cm3 and
- tii) a melt flow rate (190°C and a load of 2.16kg) of 0.1 to 50g/10 minutes,

wherein the copolymers [A-1] and [B-1] are blended at 1 to 50% and 50 to 99% by weight, respectively, based on the whole composition.

- 25. The propylene-based copolymer composition according to Claim 24, wherein ratio of density (d₂) of said ethylene-based copolymer [B-1] to density (d₁) of said ethylene/α-olefin copolymer [A-1] (d₂/d₁) is 1.05 or less.
- 26. The propylene-based-copolymer composition according to Claim 24 or 25 which has a damping factor peak caused from the glass transition temperature of said propylene/α-olefin copolymer [C-3] and that caused from the glass transition temperature of said ethylene/α-olefin copolymer composition [AB], when the temperature-dependence of the modulus of elasticity is measured, wherein these peaks are separated from each other.
 - 27. The propylene-based copolymer composition according to one of Claims 24 to 26, wherein said propylene/α-olefin copolymer composition [C-3] has a microisotacticity of 0.8 or more, with respect to the triad chains of propylene.
 - 28. The propylene-based copolymer composition according to one of Claims 24 to 27, wherein said propylene/ α -olefin copolymer [C-3] has a molecular weight distribution (Mw/Mn) of 6 or less, determined by GPC.
- 29. The propylene-based copolymer composition according to one of Claims 24 to 28, wherein said propylene/α-olefin copolymer [C-3] has a glass transition temperature Tg of -10°C or lower.
 - 30. A formed article composed of the propylene/ α -olefin copolymer composition according to one of Claims 24 to 29.
- 31. The formed article according to Claim 30 which is in the form of sheet or film.

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- 32. The formed article according to Claim 30 which is in the form of filaments.
- 33. The formed article according to Claim 30 which is in the form of medical container or tube.
- 34. The formed article according to Claim 30 which is in the form of transfusion container or tube.

INTERNATIONAL SEARCH REPORT

International application No.

PCT/JP01/04038

A. CLAS	SIFICATION OF SUBJECT MATTER		
	.Cl ⁷ C08F210/02, C08L23/08, C0	8L23/10	
	to International Patent Classification (IPC) or to both t	national classification and IPC	
	OS SEARCHED		
Minimum d Int	documentation searched (classification system followed). C1 C08F210/02, C08L23/00-23/	d by classification symbols) 3 6	
Documenta	tion searched other than minimum documentation to the	re extent that such documents are included	in the Galda and a d
Electronic o	data base consulted during the international scarch (nar	ne of data base and, where practicable, sea	rch terms used)
C. DOCU	MENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where a	ppropriate, of the relevant passages	Relevant to claim No.
Y	JP 11-209532 A (Mitsui Chemica 03 August, 1999 (03.08.99), entire description (Family: none)	ls, Ltd.),	4-34
Y	JP 11-228753 A(Mitsui Chemical 24 August, 1999 (24.08.99), entire description (Family: none)	в, Ltd.)	4-34
Y	WO 97/10295 Al (Mitsui Chemica 20 March, 1997 (20.03.97), ent & EP 791626 Al & KR 97707226 A	ls, Ltd.), ire description	4-34
Y	WO 97/10296 A1 (Mitsui Chemica 20 March, 1997 (20.03.97), ent. & EP 791627 A1 & KR 97707227 A	ls, Ltd.), ire description	4-34 .
Further	r documents are listed in the continuation of Box C.	See notest formilia annua	
"A" docume consider "E" date "L" docume cited to special docume means "P" docume than the	categories of cited documents: ent defining the general state of the art which is not red to be of particular relevance document but published on or after the international filing ent which may throw doubts on priority claim(s) or which is restablish the publication date of another citation or other reason (as specified) ent referring to an oral disclosure, use, exhibition or other ent published prior to the international filing date but later e priority date claimed	See patent family annex. To later document published after the inter priority date and not in conflict with the understand the principle or theory understand the principle or theory understand the principle or theory understand the particular relevance; the considered novel or cannot be considered step when the document is taken alone document of particular relevance; the considered to involve an inventive step combined with one or more other such combination being obvious to a person document member of the same patent fit	e application but cited to rhying the invention taken divention cannot be ed to involve an inventive taken divention cannot be when the document is documents, such skilled in the art amily
31 J	actual completion of the international search ruly, 2001 (31.07.01)	Date of mailing of the international searce 07 August, 200 (07.	
	nailing address of the ISA/ nese Patent Office	Authorized officer	
Facsimile N	0.	Telephone No.	
orm PCT/I	SA/210 (second sheet) (July 1992)		

INTERNATIONAL SEARCH REPORT

International application No.
PCT/JP01/04038

Box I Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet) This international search report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons: 1. Claims Nos.: because they relate to subject matter not required to be searched by this Authority, namely: 2. Claims Nos.: 1-3 because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically: See the extra sheet. 3. Claims Nos.: because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a). Box 12 Observations where unity of invention is lacking (Continuation of item 2 of first sheet) This International Searching Authority found multiple inventions in this international application, as follows: 1. As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims. 2. As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee. 3. As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.: 4. No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.: Remark on Protest The additional search fees were accompanied by the applicant's protest. No protest accompanied the payment of additional search fees...

Form PCT/ISA/210 (continuation of first sheet (1)) (July 1992)

INTERNATIONAL SEARCH REPORT

International application No.

PCT/JP01/04038

Continuation of Box No. I-2 of continuation of first sheet (1)

With respect to claim 1

There is no statement in the description concerning a process for producing the "ethylene copolymer [I]" of claim 1 (polymerization conditions including a catalyst system). Since the properties of a polyolefin depend on the production process, the requirements given in claim 1 alone are not sufficient to specify the "ethylene copolymer [I]."

Therefore, a meaningful international search cannot be made with respect

to the "ethylene copolymer [I] " of claim 1.

With respect to claim 2

There is no statement in the description concerning a process for producing the "ethylene copolymer satisfying the properties (b-1) to (b-30)" of claim 2 (polymerization conditions including a catalyst system). Since the properties of a polyolefin depend on the production process, the requirements given in claim 2 alone are not sufficient to specify the "ethylene copolymer satisfying the properties (b-1) to (b-30)." Furthermore, any ingredient other than the "ethylene copolymer satisfying the properties (b-1) to (b-30) contained in the ethylene copolymer composition [II] is not shown in claim 2. In addition, no explanation thereon is given in the description.

Because of this, none of the constituent components of the "ethylene copolymer composition [II]" of claim 2 can be specified. Therefore, a meaningful international search cannot be made with respect to the *ethylene copolymer

composition [II] of claim 2.

With respect to claim 3

According to claim 3, the requirements of the "propylene copolymer composition [III] " of claim 3 reside in two points, i.e., to contain a propylene copolymer and to satisfy the properties (c-1) to (c-3).

In the description, however, there is no explanation on the "propylene copolymer." Since the term propylene copolymer generally means any of a variety of propylene copolymers, the "propylene copolymer" of claim 3 is unclear. In view of the statements given in the other claims and in the description, it is proper to understand that the statements (c-1) to (c-3) are on properties of

some kind of polymer; a clerical error is highly probable.

Therefore, a meaningful international search cannot be made with respect to the "propylene copolymer composition [III] " satisfying those two requirements.

(4) Incidentally, the relationship between the "ethylene copolymer [I]," "ethylene copolymer composition [II]," and "propylene copolymer composition [III] " of claims 1 to 3 and the "copolymer [A] of ethylene and a C_{4-20} α -olefin" and "propylene copolymer composition" of other claims is unclear. Therefore, there is a possibility that this application might not comply with the requirement of unity of invention.

Form PCT/ISA/210 (extra sheet) (July 1992)